



HAL
open science

Decoding the Atomic Structure of Ga₂Te₅ Pulsed Laser Deposition Films for Memory Applications Using Diffraction and First-Principles Simulations

Andrey Tverjanovich, Chris Benmore, Maxim Khomenko, Anton Sokolov, Daniele Fontanari, Sergei Bereznev, Maria Bokova, Mohammad Kassem, Eugene Bychkov

► To cite this version:

Andrey Tverjanovich, Chris Benmore, Maxim Khomenko, Anton Sokolov, Daniele Fontanari, et al.. Decoding the Atomic Structure of Ga₂Te₅ Pulsed Laser Deposition Films for Memory Applications Using Diffraction and First-Principles Simulations. *Nanomaterials*, 2023, 13 (14), pp.2137. 10.3390/nano13142137 . hal-04217020

HAL Id: hal-04217020

<https://ulco.hal.science/hal-04217020>

Submitted on 25 Sep 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Article

Decoding the Atomic Structure of Ga₂Te₅ PLD Films for Memory Applications Using Diffraction and First-Principles Simulations

Andrey Tverjanovich ¹, Chris J. Benmore ², Maxim Khomenko ^{3,4}, Anton Sokolov ⁵, Daniele Fontanari ⁵, Sergei Bereznev ⁶, Maria Bokova ⁵, Mohammad Kassem ⁵ and Eugene Bychkov ^{5,*}

¹ Institute of Chemistry, St. Petersburg State University, 198504 St. Petersburg, Russia

² X-ray Science Division, Advanced Photon Source, Argonne National Laboratory, Argonne, Illinois 60439, USA

³ ILIT RAS-Branch of the FSRC "Crystallography and Photonics" RAS, 140700 Shatura, Moscow Region, Russia

⁴ Laboratory of Biophotonics, Tomsk State University, 634050 Tomsk, Russia

⁵ Laboratoire de Physico-Chimie de l'Atmosphère, Université du Littoral Côte d'Opale, 59140 Dunkerque, France

⁶ Department of Materials and Environmental Technology, Tallinn University of Technology, 19086 Tallinn, Estonia

* Correspondence: bychkov@univ-littoral.fr

Abstract: Neuromorphic computing, reconfigurable optical metamaterials operational over a wide spectral range, holographic and nonvolatile displays of extremely high resolution, integrated smart photonics and many other applications need phase-change materials (PCMs) of the next generation with better energy efficiency, wider temperature and spectral range for reliable operation compared to current flagship PCMs as Ge₂Sb₂Te₅ or doped Sb₂Te. Gallium tellurides are promising candidates to achieve the necessary requirements because of higher melting and crystallization temperatures, combined with a low switching power and fast switching rate. At the same time, Ga₂Te₃ and non-stoichiometric alloys appear to be atypical PCMs, characterized by regular tetrahedral structure and the absence of metavalent bonding. The sp³ gallium hybridization in cubic and amorphous Ga₂Te₅ is also different from conventional p-bonding in flagship PCMs, raising a question of the phase-change mechanism. Besides, gallium tellurides exhibit a number of unexpected and highly unusual phenomena as nanotectonic compression or viscosity anomaly just above melting. Using high-energy X-ray diffraction, supported by first-principles simulations, we will unravel the atomic structure of amorphous Ga₂Te₅ PLD films, compare it with the crystal structure of tetragonal gallium pentatelluride, and investigate electrical, optical and thermal properties of these two materials to estimate their potential for memory applications as well as for other fields.

Keywords: phase-change materials; synchrotron diffraction; first-principles molecular dynamics

1. Introduction

Brain-inspired computing [1-3], light- or electrically-controlled reconfigurable optical metamaterials over a wide spectral range from the visible to THz [4-6], holographic and nonvolatile displays of high resolution [7,8], integrated photonic circuits [9,10] and many other applications need phase-change materials (PCMs) of the next generation with better energy efficiency, wider temperature and spectral range for reliable operation compared to current flagship PCMs as Ge₂Sb₂Te₅, doped Sb₂Te, etc. Gallium tellurides seem to be promising candidates to achieve the necessary requirements because of higher melting and crystallization temperatures, combined with a low switching power and fast switching rate [11-13]. Simultaneously, Ga₂Te₃ and non-stoichiometric compositions appear to be atypical PCMs, characterized by regular tetrahedral structure, the absence of Peierls distortion in the crystalline phase and metavalent bonding [14-16].

The sp^3 gallium hybridization in cubic and amorphous Ga_2Te_3 [16,17] is also different from conventional p-bonding in flagship PCMs, raising a question of the phase-change mechanism. In addition, gallium tellurides exhibit a number of unexpected and highly unusual phenomena as nanotectonic compression [15] or viscosity anomaly just above melting [18,19]. The nanotectonic compression involves simultaneous co-crystallization of the stable ambient and metastable high-pressure (HP) forms on usual heating of glassy $g-GaTe_3$ above the glass transition temperature T_g . The appearance of metallic HP-polymorphs seems to be beneficial for the PCM performance because of higher optical and electric contrast, accompanied by lower power consumption and a possibility of multilevel writing [16]. The viscosity anomaly for Ga-Te melts in the vicinity of sesquitelluride composition [18,19] appears to be more significant than the observed “double kink” in liquid germanium tellurides and other PCM alloys [20-23], allowing to distinguish between two contrasting models: (1) fragile-to-strong transition [20-22], and incipient liquid-liquid immiscibility [23]. Recent high-energy X-ray and neutron diffraction measurements [24] have shown an enhanced small-angle scattering below the scattering vector $Q \lesssim 0.4 \text{ \AA}^{-1}$, related to dense metallic liquid droplets in a semiconducting Ga_2Te_3 melt and correlating with a non-monotonic viscosity behavior, which can be quantitatively described by the Taylor model [25] for two-phase emulsions. The incipient transient immiscibility within a semiconductor – metal transition in liquid telluride PCMs is an interesting topic for further studies.

The main goal of the current report is to unravel the atomic structure of amorphous Ga_2Te_5 PLD film using high-energy X-ray diffraction, supported by first-principles simulations, and compare it with the crystal structure of tetragonal gallium pentatelluride [26,27]. In contrast to cubic gallium sesquitelluride Ga_2Te_3 , which is a congruently melting compound in the Ga-Te binary system [28-30], tetragonal Ga_2Te_5 is stable over limited temperature range and exhibits peritectic decomposition before melting. Consequently, the relationship between amorphous material, obtained by nearly instantaneous freezing of highly excited fragments, particles, liquid globules, etc., existing in the laser-induced plasma (plume) [31], and metastable crystal is expected to be complex, leaving room for various intermediate configurations and states. A deep insight into the atomic structure and respective electronic, optical, thermal and other properties is a key for rational design of the next-gen PCMs and new functional materials for photovoltaics, thermoelectric, DNA sensing and energy storage applications [32-37].

2. Materials and Methods

2.1. Glass and Target Synthesis, Pulsed Laser Deposition

Two different synthesis strategies were applied trying to synthesize glassy $g-Ga_2Te_5$ alloys. First, a small quantity sample (300 mg) was prepared from high-purity gallium (99.999%, Neyco) and tellurium (99.999%, Cerac) in a thin-walled silica tube, sealed under vacuum (10^{-4} mbar) and placed into a rocking furnace. The maximum temperature was 1250 K. Synthesized and homogenized sample was rapidly quenched using a salty ice/water mixture. Basically, crystalline sample was synthesized using fast quenching. Second, a two-step synthesis was applied for splat quenching. Crystalline Ga_2Te_5 was prepared as the intermediate step by conventional melt quenching technique. A tiny piece of the synthesized material was placed into a silica capillary, heated to ≈ 1050 K and kept at this temperature for one hour with subsequent cooling to 950 K. After additional equilibration for 30 minutes, the sample was splat-quenched under argon atmosphere onto a fused silica plate cooled down to ≈ 80 K. Mostly vitreous sample was obtained by splat-quenching.

The gallium pentatelluride target Ga_2Te_5 for pulsed laser deposition (PLD) was synthesized in a flat-bottom silica tube of 25 mm ID. The detailed procedure was described earlier [16] giving the final monolithic target with a thickness of 7 mm.

Ga_2Te_5 thin films were deposited at room temperature by PLD onto LCD grade float glass substrates (Kintec Company) with a substrate diameter of 2” and a thickness of 1.1 mm. A Neocera PLD system equipped with a 248 nm KrF excimer laser (Coherent Compex 102 F) was used for thin film deposition. The laser beam was focused on a $\approx 5 \text{ mm}^2$ spot on the surface of the rotating target.

A detailed PLD procedure was reported previously [16,38]. The chemical composition of the PLD films was verified by energy dispersive X-ray spectroscopy and was found to be consistent with the target composition: 29.3 ± 0.6 at.% Ga and 70.7 ± 2.0 at.% Te.

2.2. XRD and DSC Characterization

In situ XRD measurements of a Ga₂Te₅ PLD film as a function of temperature have been carried out using a Rigaku Ultima IV diffractometer equipped with a Rigaku SHT-1500 high-temperature platinum camera and sealed Co K α X-ray tube. The sample was cut from the PLD film on the glass substrate and had dimensions of 9×18 mm. The heating rate was 10 K min⁻¹ in nitrogen atmosphere. The temperature range was 443 to 653 K with a step of 10 K. After temperature measurements and cooling down the sample, the diffraction pattern was recorded again under ambient conditions. Room-temperature XRD experiments have also been carried out using a Bruker D8 Advance diffractometer (the Cu K α incident radiation) equipped with a LinxEye detector.

A PLD film powder, removed from the substrate, was used for DSC measurements employing a high precision Netzsch DSC 204 F1 Phoenix instrument with μ -sensor. A standard aluminum pan was used for the experiments with a typical heating rate of 10 K min⁻¹.

2.3. Optical and Electrical Measurements

Optical absorption spectra have been measured over a wide spectral region from 700 nm to 6 μ m. A Shimadzu UV-3600 spectrophotometer was used for optical absorption measurements in the wavelength range of 700–3200 nm. A far-IR region was covered using a Bruker Tensor FTIR spectrometer, allowing the extended IR range up to 25 μ m but limited by 6000 nm because of phonon absorption in the glass substrate. The two instruments have overlapping spectral domains between 2.5 and 3.2 μ m.

The electrical conductivity of the samples was measured employing a Hewlett Packard 4194A impedance meter over the 100 Hz to 15 MHz frequency range. The sample resistance was determined using the complex impedance plots and converted into electrical conductivity using the geometrical factor. Further experimental details can be found elsewhere [15].

2.4. High-Energy X-ray Diffraction Measurements

The 6-ID-D beamline at the Advanced Photon Source (Argonne National Laboratory, Chicago) was used for high-energy X-ray diffraction measurements in the top-up mode. The photon energy was 99.9593 keV, and the wavelength was 0.124035 Å. A two-dimensional (2D) setup was used for data collection with a Varex area detector, 2880 × 2880 pixels, and a pixel size of 150 × 150 μ m². The sample-to-detector distance was 302.5 mm. A Ga₂Te₅ PLD film powder, removed from the substrate, was placed into a silica capillary, which was fixed using a sample holder of the instrument. Cerium dioxide CeO₂ was used as a calibrant. The 2D diffraction patterns were reduced using the Fit2D software [39]. The measured background intensity of the empty capillary was subtracted, and corrections were made for the different detector geometries and efficiencies, sample self-attenuation, and Compton scattering applying standard procedures [40], providing the X-ray structure factor $S_X(Q)$:

$$S_X(Q) = w_{\text{GaGa}}(Q)S_{\text{GaGa}}(Q) + w_{\text{GaTe}}(Q)S_{\text{GaTe}}(Q) + w_{\text{TeTe}}(Q)S_{\text{TeTe}}(Q) , \quad (1)$$

where $w_{ij}(Q)$ are Q -dependent X-ray weighting coefficients and $S_{ij}(Q)$ the Faber-Ziman partial structure factors.

2.5. First-Principles Simulation Details

The Born-Oppenheimer molecular dynamics implemented within the CP2K package [41] was used for modeling of the diffraction data. The generalized gradient approximation (GGA) and the PBE0 hybrid exchange–correlation functional [42,43] combining the exact Hartree–Fock and DFT approaches were used, providing better agreement with experiment [15,16,38,44–46]. The van der Waals dispersion corrections [47] were also employed, improving first-principles molecular

dynamics (FPMD) results for telluride systems [48,49]. The applied FPMD technique was similar to previous reports [44-46]. The initial atomic configurations for amorphous Ga_2Te_5 were created and optimized using the RMC_POT++ code [50] in comparison with the experimental $S_X(Q)$. The cubic simulation box, containing 210 atoms (60 Ga and 150 Te), has the size matching the experimental density. Further optimization was carried out using DFT, applying the molecularly optimized correlation consistent polarized triple-zeta valence basis set along with the norm-conserving relativistic Goedecker–Teter–Hutter-type pseudopotentials [51]. FPMD simulations were performed using a canonical NVT ensemble employing a Nosé–Hoover thermostat [52,53]. The simulation boxes were heated from 300 K to 1500 K using 100 K steps for 20–25 ps each. At the highest temperature, the systems were equilibrated for 30 ps and cooled down to 300 K using the same temperature steps but with a longer simulation time (25–30 ps). Final equilibration and data collection at 300 K were performed for 34 ps. The connectivity and ring statistics were analyzed using the R. I. N. G. S. package [54] and a modified connectivity program [55]. The pyMolDyn code [56] applying the Dirichlet–Voronoi tessellation was used for the calculation of microscopic voids and cavities.

3. Thermal Properties and Crystallization on Heating

The obtained Ga_2Te_5 PLD thin films appear to be amorphous and vitreous according to XRD and DSC measurements. Typical DSC traces of Ga_2Te_5 PLD and bulk glassy $g\text{-GaTe}_3$ are shown in Figure 1(e). The endothermic glass transition temperature T_g increases with gallium content from 448 K ($g\text{-GaTe}_3$, 25 at.% Ga) to 491 K ($g\text{-Ga}_2\text{Te}_5$, 28.57 at.%) accompanied by exothermic crystallization. Bulk $g\text{-GaTe}_3$ shows two intense thermal features, peaked at 492 and 602 K, and a very weak intermediate effect at 547 K. The 492 K peak corresponds to primary Te crystallization (both usual trigonal $P3_12$ and high-pressure monoclinic $P2_1$ forms), while the 602 K feature is giving rise to cubic ($F\bar{4}3m$) and high-pressure rhombohedral ($R\bar{3}m$) Ga_2Te_3 , suggesting the nanotectonic contraction in a viscous supercooled liquid [15].

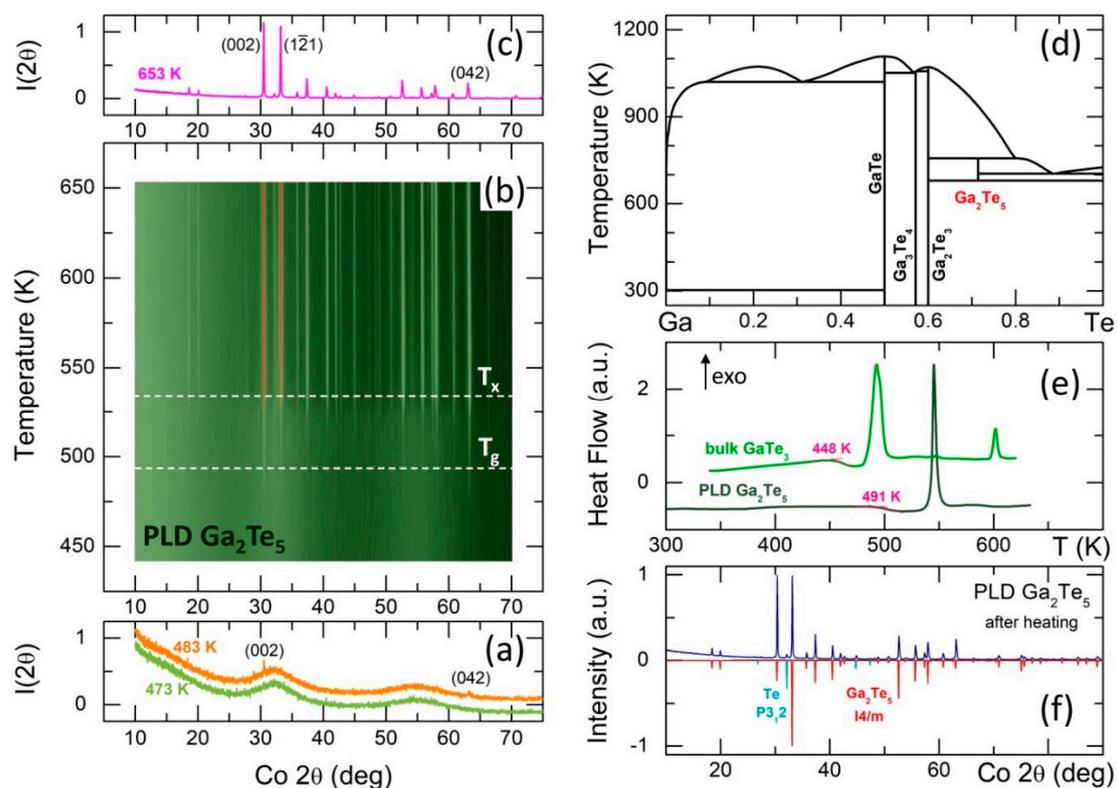


Figure 1. Thermal properties of Ga_2Te_5 PLD films and crystallization on heating. *In situ* diffraction measurements (a) at 473 and 483 K, (b) between 443 and 653 K, (c) at 653 K; (d) Ga–Te phase diagram

[28-30]; (e) typical DSC traces for Ga₂Te₅ PLD and bulk glassy g-GaTe₃, the glass transition temperatures are also indicated; (f) identification of the crystallized phases in Ga₂Te₅ PLD film after cooling from 653 K to room temperature. Trigonal tellurium, *P3₁2* [57], and tetragonal Ga₂Te₅, *I4/m* [26], were found.

In contrast, g-Ga₂Te₅ PLD film shows a single narrow and intense exothermic effect peaked at 545 K with a crystallization onset at $T_x \approx 535$ K. The *in situ* XRD measurements with a typical DSC heating rate of 10 K min⁻¹ were used to reveal the nature of crystallizing phase(s). Surprisingly, the first weak Bragg peaks at the scattering angles $2\theta = 30.5^\circ$ and 63.2° (the Co K_α incident radiation) have appeared just in the vicinity of T_g at ≈ 483 K, Figure 1(a). They correspond to the (002) and (042) reflections of tetragonal Ga₂Te₅, *I4/m* [26], which was reported to be stable only between 673 and 768 K, Figure 1(d). These unexpected results suggest the existence of certain Ga₂Te₅ motifs in the vitreous PLD film evolving into nano-crystallites on heating in a viscous supercooled liquid. Further crystallization advances in the vicinity of T_x when the remaining Bragg peaks of tetragonal Ga₂Te₅ become visible and grow over the $513 \leq T \leq 653$ K temperature range, Figure 1(b). In addition to the majority gallium pentatelluride *I4/m* polymorph, trigonal tellurium, *P3₁2* [57], was also detected as a minority phase, Figure 1(c,f). Cooling the crystallized sample, the observed phases remain intact, specifically tetragonal Ga₂Te₅, metastable at room temperature.

Even more surprisingly, gallium pentatelluride appears to be perfectly stable after 15 months at room temperature, Figure S1 (the Supplementary Information), in contrast to bulk Ga₂Te₅, transforming into cubic Ga₂Te₃ and trigonal tellurium within several weeks [28]. In other words, a controlled crystallization of the amorphous Ga₂Te₅ PLD film yields a high quality stable tetragonal crystal promising for photovoltaics, thermoelectrics, energy storage and memory applications [32-37]. On the contrary, slow cooling or fast quenching of molten Ga₂Te₅ gives a polycrystalline mixture of cubic gallium sesquitelluride and trigonal Te, Figure S1, fully consistent with the Ga-Te phase diagram, Figure 1(d).

4. Electric and Optical Properties

The measured electrical conductivity of bulk crystalline Ga₂Te₅ is shown in Figure 2(a). In contrast to previously reported results [58], the electrical conductivity follows the Arrhenius relationship over the entire temperature range

$$\sigma = \sigma_0 \exp(-E_a/k_B T), \quad (2)$$

where σ_0 is the pre-exponential factor, E_a the conductivity activation energy, k_B and T have the usual meaning. Nevertheless, the derived activation energy $E_a = 0.227 \pm 0.003$ eV is identical to that for intrinsic conductivity in tetragonal single-crystal Ga₂Te₅, measured in the direction perpendicular to the *c*-axis [58]. The conductivity pre-factor, $\sigma_0 = 17 \pm 3$ S cm⁻¹, is located at the lower limit of electronic transport regime over the extended states [59], that is, $2E_a = 0.45$ eV is approximately the electrical bandgap.

The electrical conductivity for glassy g-Ga₂Te₅ was obtained by interpolation of the available data for amorphous and glassy Ga_{*x*}Te_{1-*x*} alloys, $0 \leq x \leq 0.4$ [15,60,61], Figure 2(b). The room-temperature conductivity appears to be a decreasing exponential function of the gallium content x ,

$$\sigma_{298}(x) = \sigma_{298}(0) \exp(ax), \quad (3)$$

where $\sigma_{298}(0)$ is the conductivity of amorphous Te, and $a < 0$ is a constant. In other words, the electronic conductivity of disordered Ga-Te materials essentially depends on tellurium concentration. The interpolated g-Ga₂Te₅ conductivity is lower by two orders of magnitude than σ_{298} of bulk crystalline pentatelluride. The estimated conductivity activation energy, $E_a = 0.41$ eV, and the pre-exponential factor, $\sigma_0 \approx 220$ S cm⁻¹, suggest the electrical bandgap of glassy polymorph is about 0.8 eV.

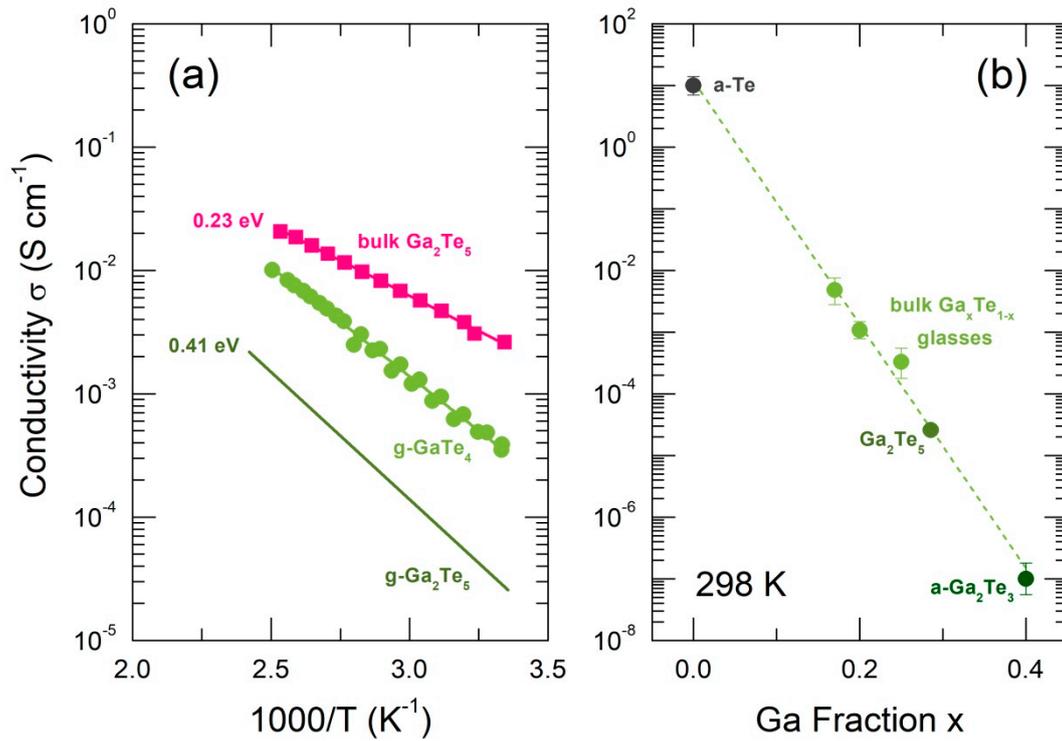


Figure 2. Electrical properties of $\text{Ga}_x\text{Te}_{1-x}$ amorphous, glassy and crystalline materials: (a) electrical conductivity of bulk crystalline Ga_2Te_5 (this work), bulk glassy g-GaTe_4 [15], and interpolated conductivity for $\text{g-Ga}_2\text{Te}_5$, the derived activation energies E_a are also indicated; (b) room-temperature conductivity for amorphous a-Te thin film, corresponding to band-to-band electronic transport [60], bulk Ga-Te glasses [15], and amorphous $\text{a-Ga}_2\text{Te}_3$ [61]; the interpolated value for $\text{g-Ga}_2\text{Te}_5$ is also shown.

Figure 3 shows the optical properties of Ga_2Te_5 PLD films. The absorption measurements reveal the fundamental optical absorption edge below the incident wavelength $\lambda \lesssim 1.2 \mu\text{m}$ accompanied by distinct interference fringes indicating a homogeneous nature and uniform thickness of the PLD film. The observed fringes allow both the refractive index n_R and the film thickness to be calculated using the Swanepoel method [62]. In addition, the well-defined interference over the used spectral range and rather thick PLD film ($2.7 \mu\text{m}$) enable also the refractive index dispersion $n_R(\lambda)$ to be estimated.

Two approaches are usually applied to represent the refractive index dispersion [63]. The Cauchy approximation of the derived data $n_R(\lambda)$ is given in Figure 3(c)

$$n_R(\lambda) = A + \frac{B}{\lambda^2} + \frac{C}{\lambda^4}, \quad (4)$$

where A , B , and C are constants which are characteristic of any given material. Since the Cauchy equation is inappropriate in a region of anomalous dispersion [63], the Sellmeier approach is often used, considering the existence in an optical material of dipole oscillators with a resonance frequency ν_0 :

$$n_R(\lambda)^2 = 1 + \frac{A\lambda^2}{\lambda^2 - \lambda_0^2}, \quad (5)$$

where A and $\lambda_0 = c/\nu_0$ are two characteristic constants, and c is the speed of light. Usually, the Sellmeier equation is written with a series of terms to account for different resonance frequencies over an extended domain, that is, ν_0 , ν_1 , etc.

$$n_R(\lambda)^2 = 1 + \frac{A_0\lambda^2}{\lambda^2 - \lambda_0^2} + \frac{A_1\lambda^2}{\lambda^2 - \lambda_1^2} + \dots \quad (6)$$

The Sellmeier coefficients A_i and λ_i allow the normal dispersion of optical glasses to be calculated over a wide spectral range. In our case, we were limited to the original Sellmeier equation (5) with the following coefficients: $A_0 = 3.5017$ and $\lambda_0 = 0.3992$, Figure 3(d). The higher order terms

were inaccessible caused by insufficient spectral range and experimental uncertainty. Nevertheless, the two approaches describe reasonably well the derived $n_R(\lambda)$ values.

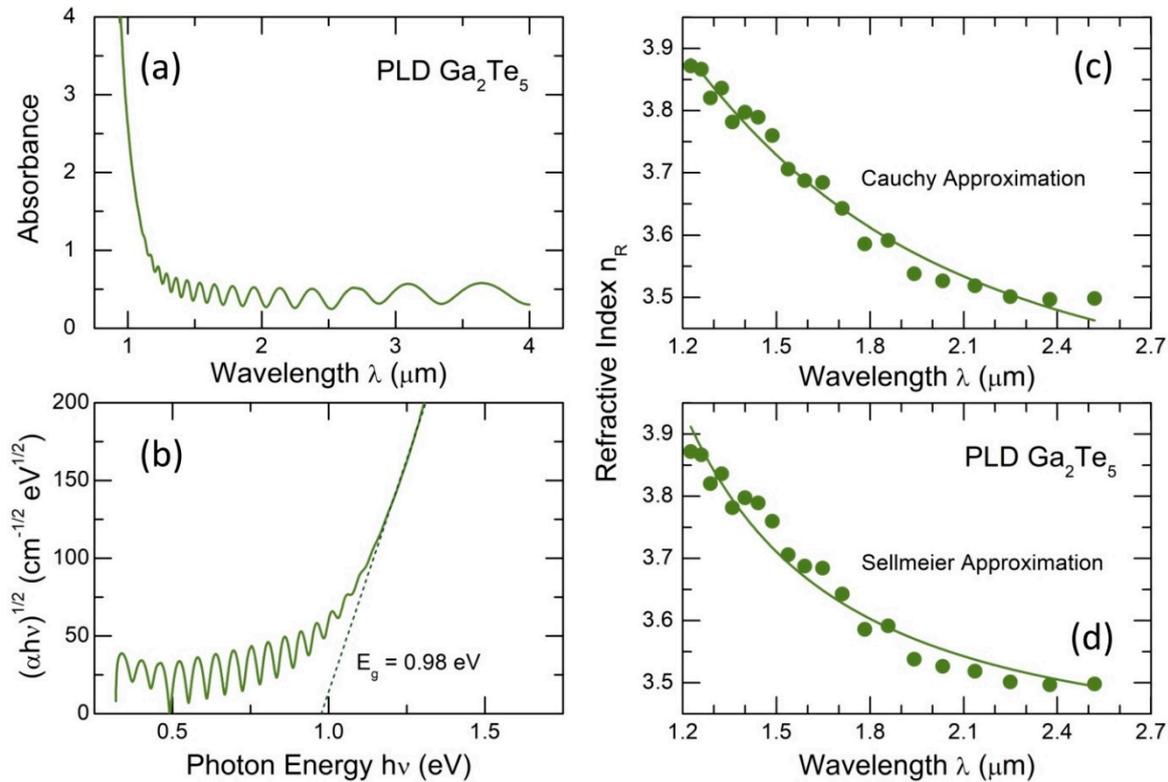


Figure 3. Optical properties of Ga₂Te₅ PLD films: (a) optical absorbance with interference fringes as a function of the incident wavelength λ ; (b) the Tauc plot, $\sqrt{\alpha h\nu}$ vs. photon energy $h\nu$, where α is the absorption coefficient, yielding the optical band gap E_g ; (c) the Cauchy and (d) Sellmeier approximations [63] describing the dispersion of the refractive index n_R as a function of λ . See the text for further details.

The optical absorption results were also used to calculate the optical bandgap E_g applying the Tauc relation [64]:

$$\alpha = \frac{A(h\nu - E_g)^2}{h\nu}, \quad (7)$$

where α is the absorption coefficient, $h\nu$ the photon energy, and $A \approx 10^5 \text{ cm}^{-1} \text{ eV}^{-1}$ is a constant.

As expected, the derived bandgap $E_g = 0.98 \pm 0.02 \text{ eV}$, Figure 3(b), for glassy gallium pentatelluride was found to be smaller than that for g-Ga₂Te₃, 1.20 eV [16], supporting a predominant role of the tellurium content on electronic and optical properties of Ga-Te alloys. Simultaneously, the optical E_g appears to be comparable with the electrical counterpart, $2E_a = 0.82 \text{ eV}$.

Thermally annealed and crystallized Ga₂Te₅ PLD film exhibits more complicated optical absorption, Figure 4. The absorbance below $\lambda \lesssim 1 \mu\text{m}$ shows a distinct blue-shift, while the low-energy absorbance becomes more intense and mostly loses interference fringes indicating less homogeneous material in both the chemical composition and thickness. Taking into account the existence of (nano)crystallites in the annealed PLD film, additional scattering corrections were applied simultaneously with the usual reflection corrections. The refractive index $n_R(\lambda)$ of g-Ga₂Te₅ was used for these calculations.

The Mie theory of light scattering for turbidity τ measurements and the wavelength exponent $\lambda^{-\chi}$ were employed for the scattering corrections [65,66]

$$\tau = l^{-1} \ln(I_0/I), \quad (8)$$

where l is the scattering path length, and I_0 and I are the intensities of the incident and transmitted beam, respectively. The turbidity depends on several parameters

$$\tau = B_s N_p V_p^2 \lambda^{-\chi}, \quad (9)$$

where B_s is the scattering coefficient, N_p the particle number density, V_p the average volume of the particle, and χ is the wavelength exponent. Combining Eqs. (8) and (9), one obtains

$$D = K_s \lambda^{-\chi}, \quad (10)$$

where $D = \log(I_0/I)$ is the optical density, and K_s is a constant. The values of K_s and χ were obtained plotting $\ln D$ vs. $\ln \lambda$ (Figure S2), which allows both the turbidity $\tau(\lambda)$ and χ to be determined and the scattering corrections to be calculated. The theoretical Heller wavelength exponent χ_0 [65] yields the average particle size $\langle r_p \rangle$, which appears to be $\langle r_p \rangle \approx 110$ nm for c-Ga₂Te₅, Figure S2. The derived $\langle r_p \rangle$ value is consistent with the size of crystallites, obtained from the XRD linewidth Γ , yielding $\langle r_p^{\text{XRD}} \rangle = 2\pi/\Gamma > 50$ nm. The final absorbance corrected for reflection and scattering is shown in Figure 4(a).

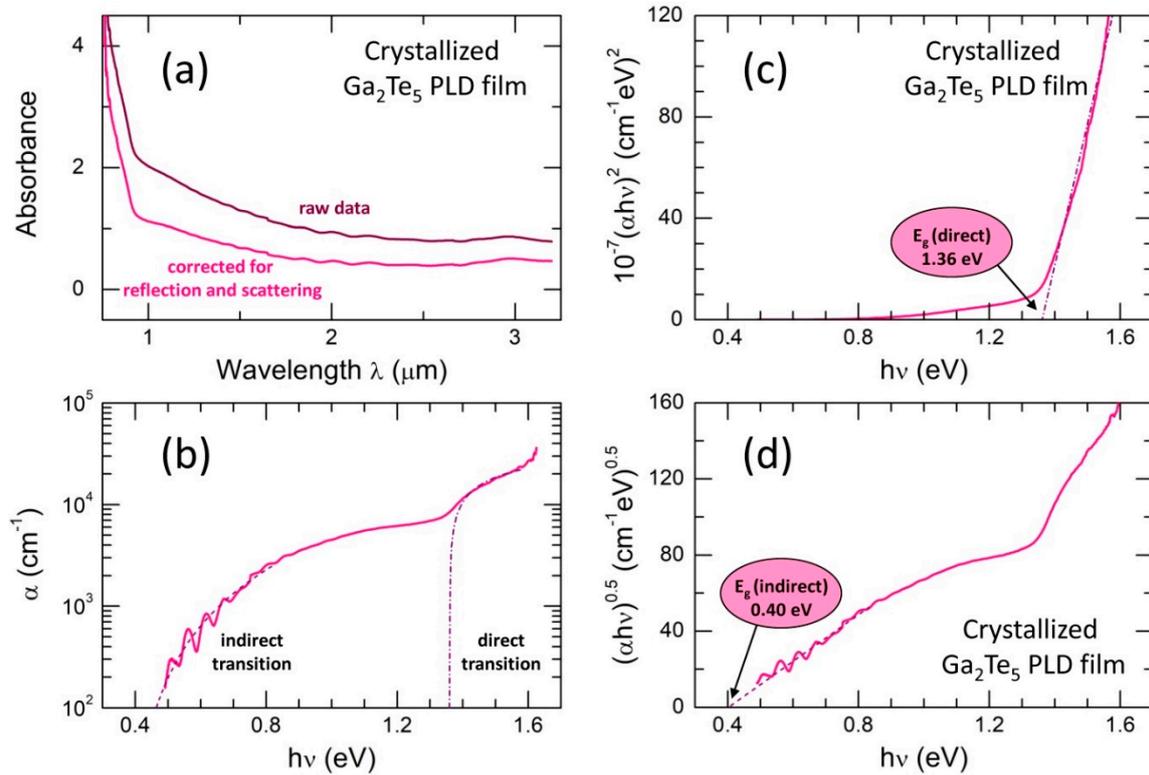


Figure 4. Optical properties of crystallized Ga₂Te₅ PLD film: (a) raw absorbance data and absorbance corrected for reflection and scattering; (b) absorption coefficient α and the analysis results for direct (the dash-dotted line) and indirect (the dashed line) optical transitions; (c) data analysis suggesting direct optical transition; the derived direct optical gap E_g^d is also indicated; (d) data analysis assuming indirect optical transition; the derived indirect optical gap E_g^i is indicated. See the text for further details.

The derived optical absorption coefficient α , presented in Figure 4(b), exhibits two optical processes above and below the incident photon energy $h\nu \approx 1.3$ eV. Basically, the overall shape of α is reminiscent of that in silicon or carbon [67,68], which is related to direct optical transition at high $h\nu$ and indirect optical absorption at low photon energies [68-72]. Assuming direct optical transition in crystallized Ga₂Te₅ above 1.3 eV

$$ah\nu = A(h\nu - E_g^d)^{1/2}, \quad (11)$$

the direct optical bandgap was found to be $E_g^d = 1.36 \pm 0.03$ eV, Figure 4(c). The constant A in Eq (11) is given by $A \approx [e^2/(n_R ch^2 m_e^*)](2m^*)^{3/2}$, where e is the electron charge, m^* is a reduced electron and hole effective mass [69].

The optical absorption plotted as $\sqrt{ah\nu}$ vs. photon energy $h\nu$, Eq. (7), yields the indirect optical bandgap $E_g^i(1) = 0.40 \pm 0.03$ eV, Figure 4(d). The derived value is consistent with the electrical bandgap of c-Ga₂Te₅, $2E_a = 0.45$ eV, Figure 2(a).

The experimental data for tetragonal Ga₂Te₅, obtained using the conductivity and Hall effect measurements, are strongly anisotropic and changing over a wide range between 0.46 and 1.79 eV [58]. The calculated E_g values are also variable, $0.86 \leq E_g \leq 1.7$ [72,73], depending on the applied simulation method. Nevertheless, the results of electrical and optical measurements show a reasonable contrast between amorphous (SET) and crystalline (RESET) states for Ga₂Te₅.

5. High-Energy X-Ray Diffraction

The high-energy X-ray diffraction data in Q -space are shown in Figure 5. In contrast to fast cooled Ga₂Te₅ in a thin-walled silica tube, mostly consisting of cubic Ga₂Te₃ and trigonal tellurium with some vitreous fraction (Figure S1), the splat quenching of tiny Ga₂Te₅ droplets yields essentially glassy material with non-negligible nanocrystals of cubic gallium sesquisulfide, Figure 5(a). The spontaneous Ga₂Te₃ crystallization is consistent with the Ga-Te phase diagram, Figure 1(d), related to peritectic decomposition of Ga₂Te₅ above 768 K. The obtained Ga₂Te₅ PLD films are fully vitreous with a distinct glass transition temperature at 491 K, Figure 1(e). The X-ray structure factor $S_X(Q)$ of g-Ga₂Te₅ PLD appears to be intermediate between bulk glassy GaTe₄ and Ga₂Te₃ PLD film, Figure 5(b), suggesting structural similarities and revealing a systematic evolution of vitreous Ga_xTe_{1-x} materials with increasing gallium content x independently on preparation techniques.

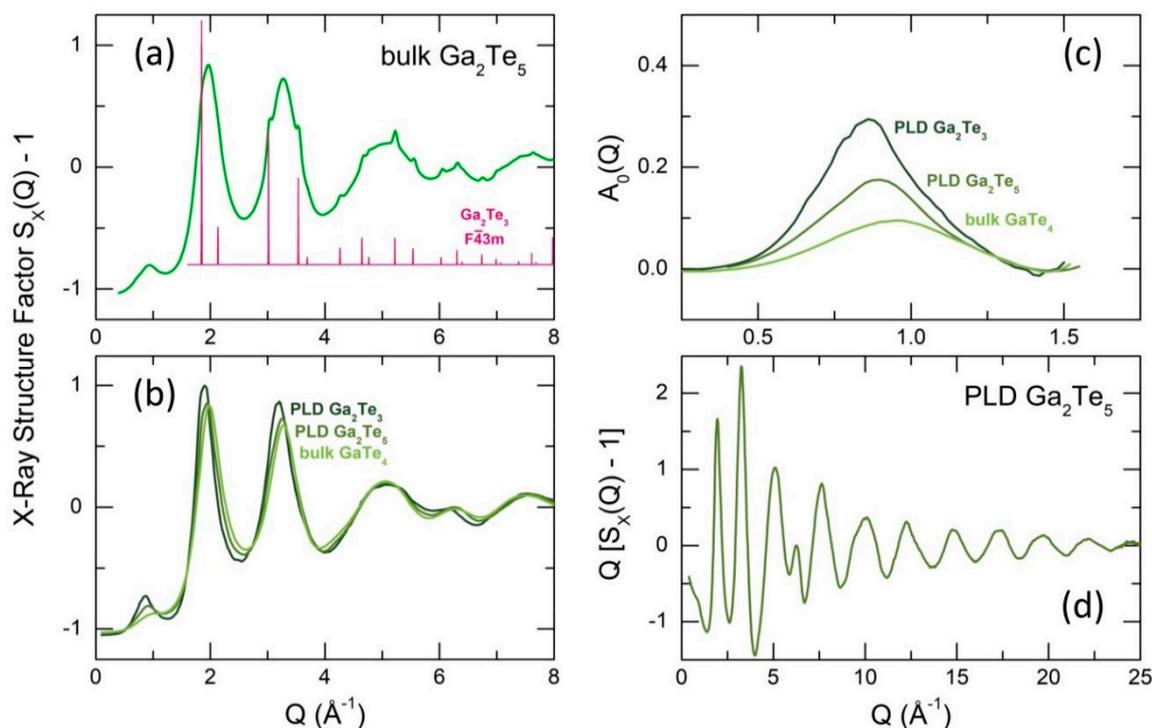


Figure 5. Diffraction data in Q -space: the X-ray structure factor $S_X(Q)$ of (a) bulk splat-quenched Ga₂Te₅ and (b) Ga₂Te₃, Ga₂Te₅ PLD films and bulk glassy g-GaTe₄ over a limited Q -range; (c) isolated first sharp diffraction peaks (FSDP) for Ga₂Te₃ and Ga₂Te₅ PLD films, and g-GaTe₄; (d) the interference function $Q[S_X(Q)-1]$ for Ga₂Te₅ PLD film over the extended Q -range. The Bragg peaks for cubic Ga₂Te₃ (space group $F\bar{4}3m$) are also shown in (a).

In particular, we observe an emerging and growing first sharp diffraction peak (FSDP), also shifting to lower Q with increasing x from $Q_0 = 0.94 \pm 0.01 \text{ \AA}^{-1}$ (GaTe₄, $x = 0.2$) to $0.86 \pm 0.01 \text{ \AA}^{-1}$ (Ga₂Te₃, $x = 0.4$), Figure 5(c). The isolated FSDPs were obtained using the subtraction procedure [74,75]. The FSDP systematics (position Q_0 and amplitude A_0) reveals monotonic nearly linear trends as a function of x , Figure S3, for both bulk glasses ($0.17 \leq x \leq 0.25$) and vitreous PLD films ($0.2857 \leq x \leq 0.40$), suggesting structural similarities on the short- and intermediate-range scale.

Distinct high- Q oscillations, clearly visible over the extended Q -range for the Ga_2Te_5 interference function $Q[S_X(Q)-1]$, Figure 5(d), enable high real-space resolution for atomic pair-distribution $g_X(r)$ and total correlation $T_X(r)$ functions after usual Fourier transform

$$T_X(r) = 4\pi\rho_0r + \frac{2}{\pi} \int_0^{Q_{\max}} Q[S_X(Q) - 1] \sin QrM(Q)dQ, \quad (12)$$

where ρ_0 is the experimental number density, $M(Q)$ the Lorch window function, and $Q_{\max} = 30 \text{ \AA}^{-1}$.

The derived $T_X(r)$ for g- Ga_2Te_5 PLD film is shown in Figure 6. The asymmetric feature between 2.4 and 3.2 \AA corresponds to Ga-Te and Te-Te nearest neighbors (NN). A Gaussian fitting (Table 1) yields tetrahedral gallium coordination at 2.64 \AA , consistent with the Ga-Te coordination numbers and NN distances in crystalline and glassy gallium tellurides [14-16,26-28,77,78]. On the contrary, the Te-Te atomic pairs in glassy Ga_2Te_5 are markedly shorter (2.80 \AA) than those in tetragonal gallium pentatelluride (3.027 \AA) [26]. Nevertheless, this Te-Te NN distance is typical for amorphous and trigonal tellurium [57,79] and Te-rich binary and ternary glasses [15,77,78,80]. The partial $T_{ij}(r)$ correlation functions for tetragonal gallium pentatelluride are compared in Figure 6(b) with experimental $T_X(r)$ for g- Ga_2Te_5 . We note both similarities and differences for the two materials.

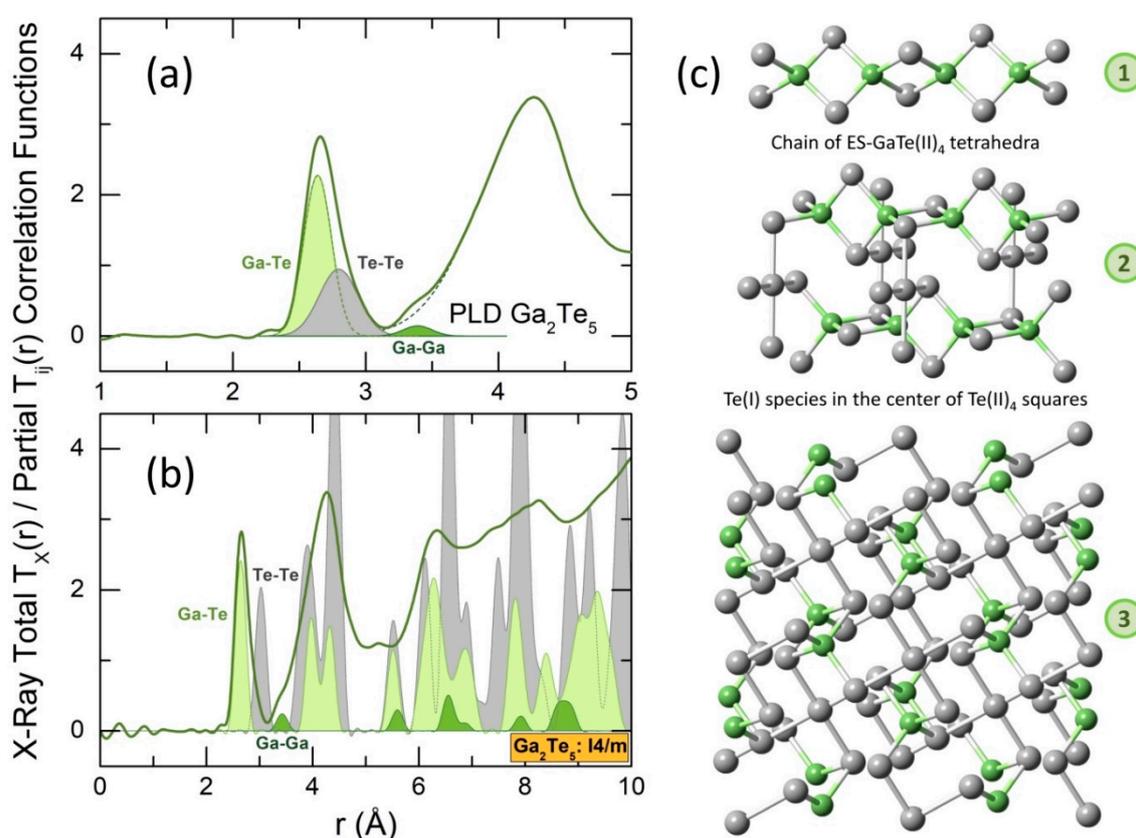


Figure 6. Diffraction data in r -space: the X-ray total correlation function $T_X(r)$ for g- Ga_2Te_5 PLD film, showing (a) a four-peak Gaussian fitting of the nearest (NN) and second (2N) neighbor features between 2.4 and 4.5 \AA ; the Ga-Te and Te-Te NNs are highlighted in light green and gray, respectively; the Ga-Ga 2Ns are green; (b) a comparison with the partial correlation functions $T_{ij}(r)$ for tetragonal Ga_2Te_5 (space group $I4/m$ [26]), derived using the XTAL code [76]; (c) the crystal structure of tetragonal gallium pentatelluride, revealing (1) infinite chain of edge-sharing ES- GaTe(II)_4 tetrahedra along the c -axis, (2) two neighboring chains connected by Te(I) species located in the center of Te(II) squares, (3) an approximate (a, b) projection of the crystal structure.

Table 1. Interatomic distances r_{ij} and partial coordination numbers N_{ij} of the nearest neighbors in glassy Ga_2Te_5 PLD films according to high-energy X-ray diffraction and first-principles molecular dynamics.

Ga-Ga		Ga-Te		Te-Te		$N_{\text{Ga-X}}$
r_{ij} (Å)	N_{ij}	r_{ij} (Å)	N_{ij}	r_{ij} (Å)	N_{ij}	
High-energy X-ray diffraction						
–	–	2.637(3)	3.98(5)	2.796(5)	1.01(8)	3.98(5)
First-principles molecular dynamics (GGA/PBE0)						
2.417	0.03	2.615*	3.97	2.802*	0.93	4.00

* asymmetric peak.

The crystalline counterpart is stable over a narrow temperature range from 673 to 768 K [28-30], Figure 1(d). In contrast to layered Al_2Te_5 and In_2Te_5 [27], gallium pentatelluride has a 3D structure consisting of infinite chains, parallel to the c axis, formed by edge-sharing ES-GaTe(II)_4 tetrahedra, Figure 6(c). Every four ES-GaTe(II)_4 entities, belonging to the neighboring chains, are linked together by Te(I) species, which are located in the center of squares, formed by Te(II).

These square-planar Te_5 units (crosses) are presumably mostly missing in glassy Ga_2Te_5 since the Te-Te NN coordination number is significantly lower in the glass, $N_{\text{Te-Te}}^{\text{PLD}} = 1.01 \pm 0.08$ (Table 1), compared to that in the crystal, $N_{\text{Te-Te}}^{14/m} = 1.6 = \frac{1}{5} \times 4 + \frac{4}{5} \times 1$; the average Te-Te coordination for the Te(I) and Te(II) species, Figure 6(c). The ES-GaTe_4 tetrahedra seem existing in the glass network, since a weak shoulder at 3.39 ± 0.02 Å is observed for the asymmetric second neighbor peak, centered at ≈ 4.3 Å. The short Ga-Ga second neighbor correlations, characteristic of ES-units in tetragonal Ga_2Te_5 , are located at 3.424 Å, Figure 6(b). However, a weak average Ga-Ga weighting factor, $\langle w_{\text{GaGa}}^{\text{X}}(Q) \rangle = 0.02964$ vs. $\langle w_{\text{GaTe}}^{\text{X}}(Q) \rangle = 0.28405$ or $\langle w_{\text{TeTe}}^{\text{X}}(Q) \rangle = 0.68630$, and, consequently, a small amplitude of this feature as well as the truncation ripples, related to a finite Q -range of the Fourier transform, enable only a rough estimation of $N_{\text{Ga-Ga}}^{\text{PLD}} = 1.2 \pm 0.4$, compared to $N_{\text{Ga-Ga}}^{14/m} = 2$. A deep insight into the atomic structure of vitreous Ga_2Te_5 PLD films yields first-principles molecular dynamics.

6. First-Principles Molecular Dynamics

Simulated FPMD X-ray structure factor $S_{\text{X}}(Q)$ and pair-distribution function $g_{\text{X}}(r)$ for glassy Ga_2Te_5 in comparison with experimental results are shown in Figure 7(a,b). We note that the GGA approximation with hybrid PBE0 functional describes well the experimental data as it was reported earlier [15,16,44-46]. The positions and amplitudes of the diffraction features in both Q - and r -space are reproduced.

The calculated partial structure factors $S_{ij}(Q)$ are displayed in Figure 7(c). As expected, the main contribution to the FSDP comes from the Ga-Ga partial $S_{\text{GaGa}}(Q)$. The simulated $g_{ij}(r)$, Figure 7(d), reveal complicated short- and intermediate-range order.

The asymmetric Ga-Te NN correlations are peaked at 2.62 Å and suggest at least two contributions with slightly different bond lengths. The Ga-Te coordination number is consistent with the experiment, $N_{\text{Ga-Te}}^{\text{FPMD}} = 3.97$, Table 1, assuming a tetrahedral gallium local environment. In addition to homopolar Te-Te bonds at 2.80 Å, a weak Ga-Ga NN feature at 2.42 Å was also found. The amplitude of this peak is too small to be observed experimentally, Figure 7(b). The Ga-Ga second neighbors between 3 and 4.5 Å have a bimodal distribution. The shoulder at ≈ 3.35 Å indicates the ES-units, while the main contribution at 3.92 Å is related to corner-sharing CS-entities. Consequently, the fraction of ES-GaTe_4 , $f_{\text{ES}}^{\text{FPMD}} = 0.45$, is significantly lower than that in tetragonal polymorph, $f_{\text{ES}}^{14/m} = 1$, that is, only ES-GaTe_4 are present. The experimental value, $f_{\text{ES}}^{\text{PLD}} = 0.6 \pm 0.2$, is reasonably consistent with the FPMD result. Basically, the experimental and FPMD structural parameters were found to be similar or identical, Table 1.

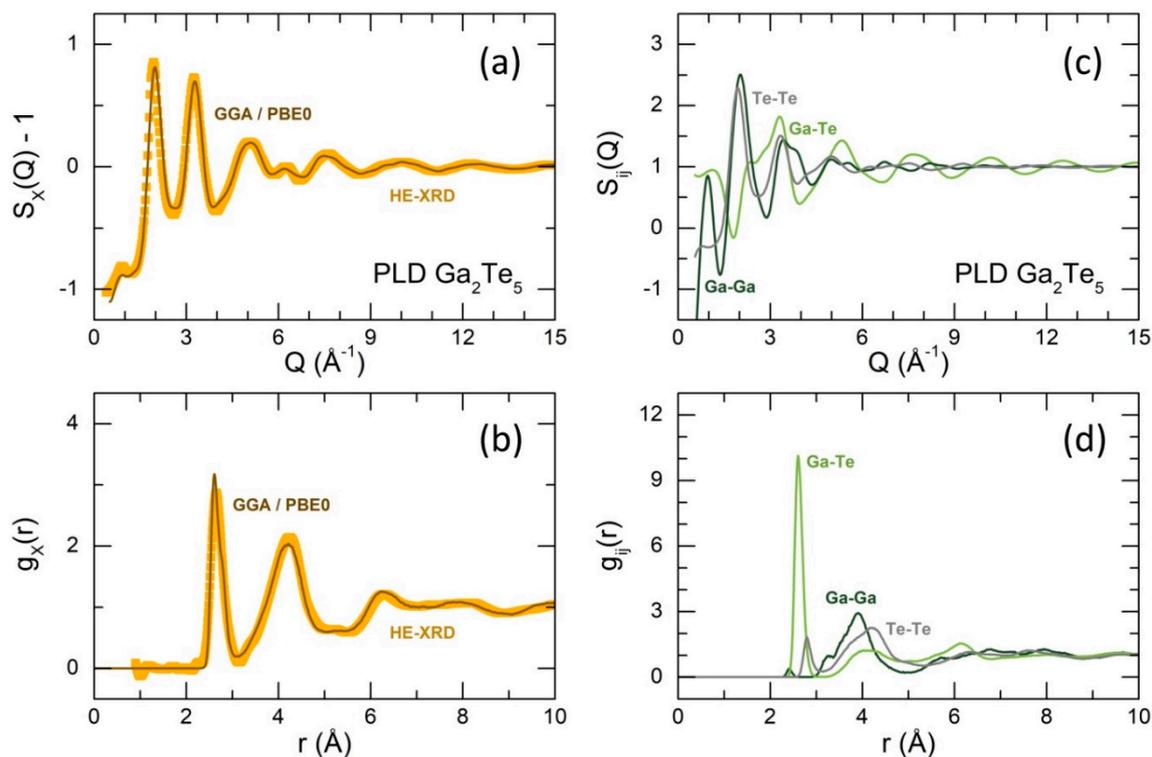


Figure 7. First-principles molecular dynamics modeling of Ga_2Te_5 PLD film using GGA/PBE0 approximation; calculated and experimental (a) X-ray structure factors $S_X(Q)$, (b) pair-distribution functions $g_X(r)$, (c) partial structure factors $S_{ij}(Q)$, (d) partial pair-distribution functions $g_{ij}(r)$. The Ga-Ga, Ga-Te and Te-Te partials are dark green, light green and grey, respectively.

The Ga and Te local coordination distributions are presented in Figure 8. The tetrahedral gallium coordination $\text{Ga}(\text{Te}_{4-m}\text{Ga}_m)$ contains negligible number of Ga-Ga homopolar pairs ($m = 1$). In contrast, tellurium has multiple coordination environments $\text{Te}(\text{Ga}_{n-m}\text{Te}_m)$, $1 \leq n \leq 4$, but only two-fold $\text{Te}_{2\text{F}}$ (50.5%) and three-fold $\text{Te}_{3\text{F}}$ (47.6%) coordinated species appear to be abundant. The tellurium forms reveal a large variability in Te-Te bonds, $0 \leq m \leq 4$, from pure heteropolar Te-Ga coordination ($m = 0$) to fully homopolar environment ($m = n$). We should, however, note a small fraction of $\text{Te}_{4\text{F}}$ species (1.87%) and a negligible number of Te_5 square-planar entities (0.23%), the only form of tellurium subnetwork in tetragonal gallium pentatelluride, Figure 6(c). This result is coherent with the reduced Te-Te coordination number $N_{\text{Te-Te}}^{\text{PLD}} \cong N_{\text{Te-Te}}^{\text{FPM}} \cong 1$, Table 1.

The geometry of GaTe_4 units yields either Te-Ga-Te bond angles or the orientational order parameter q [81,82]. Figure 9(a,c) shows the calculated $B_{\text{TeGaTe}}(\theta)$ bond angle distribution for $g\text{-Ga}_2\text{Te}_5$ in comparison with tetragonal Ga_2Te_5 and cubic Ga_2Te_3 , respectively. We note a broad and slightly asymmetric $B_{\text{TeGaTe}}(\theta)$ function, centered at $103.3 \pm 0.3^\circ$, for the PLD film (Figure S4). The Te-Ga-Te angles in the two crystalline references, characterizing both distorted ES- GaTe_4 tetrahedra in gallium pentatelluride and regular CS-units in cubic sesquitelluride, are located within the glassy envelope but not reproducing it by simple broadening. Nevertheless, the tetrahedral geometry in tetragonal Ga_2Te_5 seems to be closer to that in the glass.

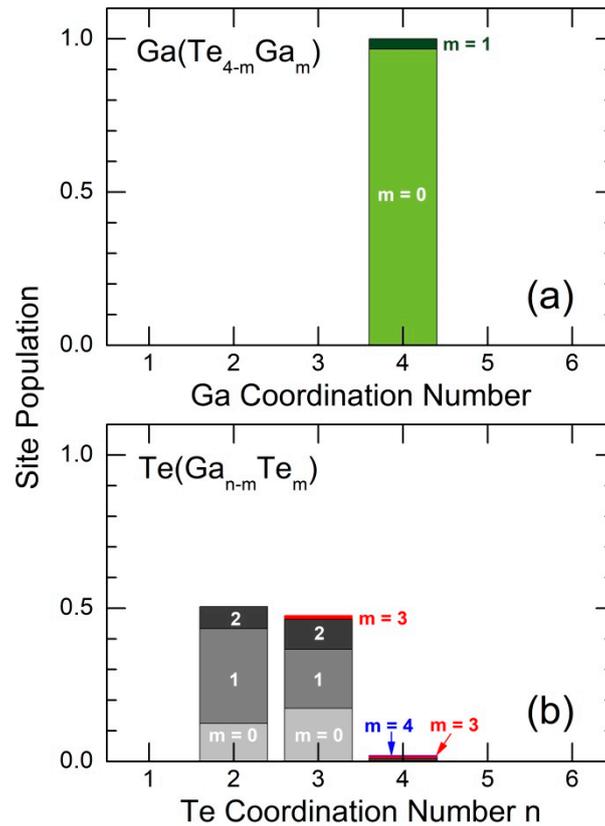


Figure 8. (a) Gallium and (b) tellurium coordination distributions. Tetrahedral gallium coordination $\text{Ga}(\text{Te}_{4-m}\text{Ga}_m)$ contains negligible number of Ga-Ga homopolar pairs ($m = 1$). Multiple tellurium coordination environments $\text{Te}(\text{Ga}_{n-m}\text{Te}_m)$, $1 \leq n \leq 4$, contain a significant number of Te-Te bonds, $0 \leq m \leq 4$. See the text for further details.

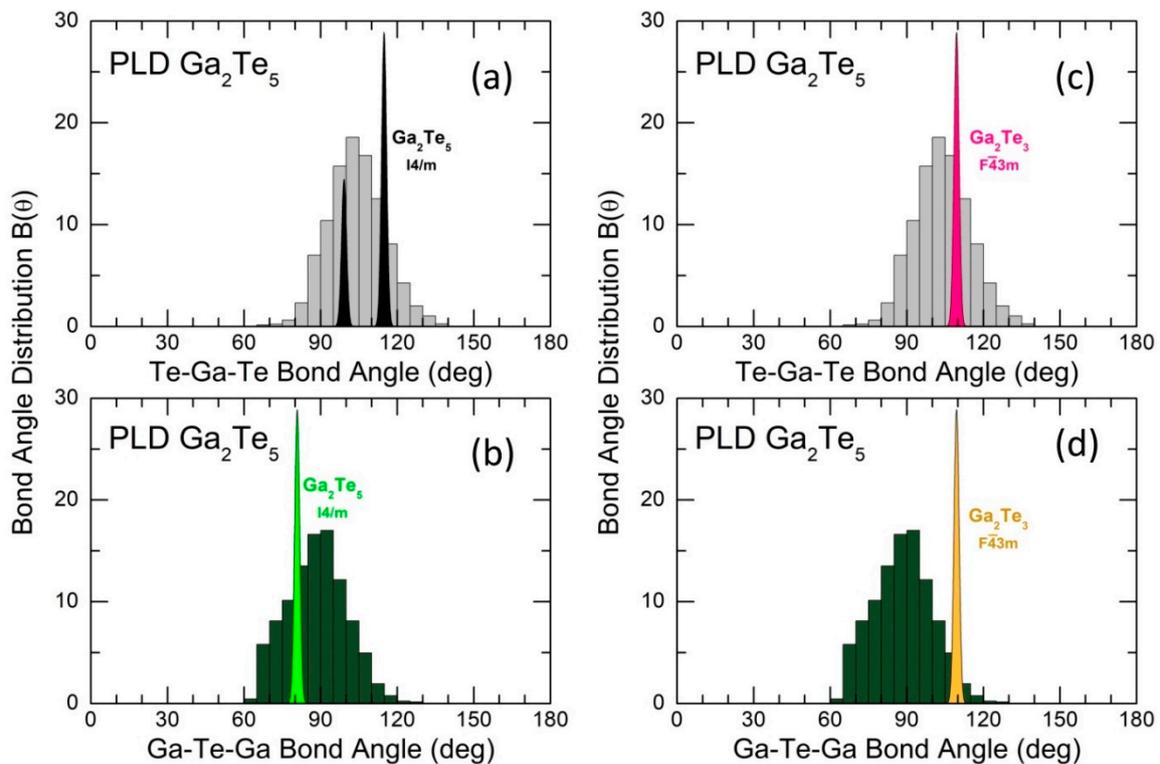


Figure 9. Bond angle distributions $B(\theta)$ in simulated g- Ga_2Te_5 : (a,c) the Te-Ga-Te and (b,d) Ga-Te-Ga bond angles in comparison with (a,b) tetragonal Ga_2Te_5 and (c,d) cubic Ga_2Te_3 .

The connectivity of GaTe_4 tetrahedra is given by the Ga-Te-Ga triplets or the respective $B_{\text{GaTeGa}}(\theta)$ distributions, Figure 9(b,d). A difference to the crystalline connectivity in this case is even more significant but a remote resemblance to connected ES-entities in the tetragonal crystal also exists.

The order parameter q [81,82] is often used to evaluate the polyhedral topology and distinguish between tetrahedral and non-tetrahedral local geometry of four-fold coordinated GaTe_4 entities

$$q = 1 - \frac{3}{8} \sum_{j=1}^3 \sum_{k=j+1}^4 \left(\cos \psi_{jk} + \frac{1}{3} \right)^2, \quad (13)$$

where ψ_{jk} is the Te-Ga-Te angle of the central gallium atom with its nearest Te neighbors j and k . The average value of q changes between 0 for an ideal gas and $q = 1$ for a regular tetrahedral network. The $P(q)$ probability distribution function is shown in Figure 10(a). Asymmetric $P(q)$ is peaked at $q = 0.93$ and decreases sharply both ways to high and low q . Usually, the tetrahedral limit is set at $q \geq 0.8$ [44,83]. The $P(q)$ integration within these limitations shows that 97% of GaTe_4 units belong to tetrahedral geometry. The remaining entities (3%) presumably are defect octahedral species GaTe_4 with two missing Te neighbors around the central Ga atom, characterized by $0.4 < q < 0.8$. The regular defect octahedron ($\psi_1 = \pi$ and $\psi_2 = \pi/2$) has $q = 5/8$.

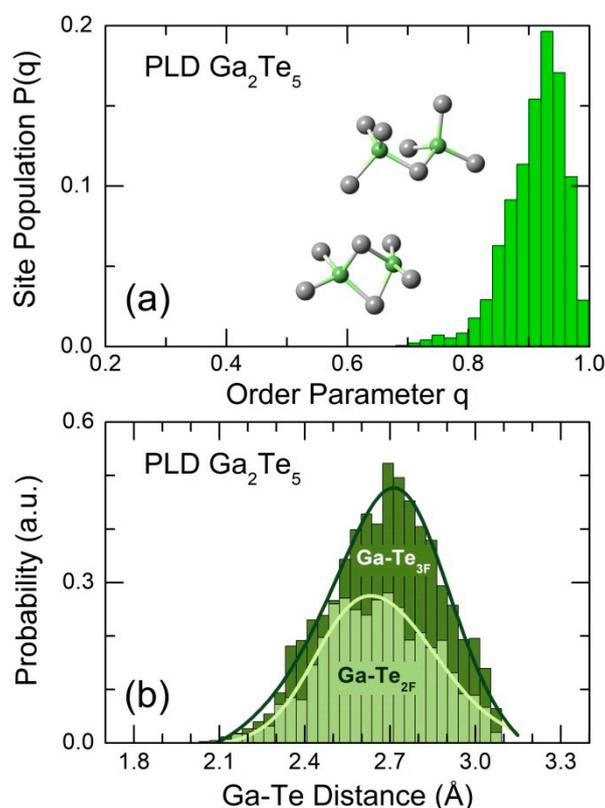


Figure 10. (a) Orientational order parameter q [81,82] for 4-fold coordinated Ga species in $g\text{-Ga}_2\text{Te}_5$, and (b) Ga-Te NN distance distributions for two-fold Te_{2F} and three-fold Te_{3F} coordinated tellurium atoms.

Two-fold and three-fold coordinated tellurium can explain the asymmetric shape of the Ga-Te NN peak. The calculated Ga-Te_{2F} and Ga-Te_{3F} distances are presented in Figure 10(b). The two distributions are broad and asymmetric but have slightly different maxima. The Ga-Te_{3F} bonds are longer (a broad maximum at 2.71 ± 0.03 Å) compared to Ga-Te_{2F} , peaked at 2.63 ± 0.02 Å. Similar difference, $\Delta r_{3F-2F} = 0.12 \pm 0.01$ Å, was reported in monoclinic Ga_2S_3 [84] with ordered distribution of gallium vacancies. The ratio of Ga-Te_{3F} to Ga-Te_{2F} bond populations, $r = [\text{Ga-Te}_{3F}]/[\text{Ga-Te}_{2F}] = 1.75$, was also found to be similar to the expected stoichiometric ratio $r_0 = 2$ for tetrahedral Ga species, having the formal oxidation state Ga^{3+} and Ga_2Te_3 alloy composition.

The connectivity analysis shows that all Ga and Te species are connected. The analysis of Te-Te connectivity reveals a different size of Te_k fragments, Figure 11(a). Tellurium monomers ($k = 1$), that is, Te atoms with only heteropolar Te-Ga bonds, and dimers ($k = 2$) represent a relative majority, 55%, of all Te_k fragments. The remaining fragments can be divided into two groups: (i) $3 \leq k \leq 6$ (see the inset in Figure 11(a)), and (ii) oligomeric chains, $k = 15$ for the used 210-atom simulation box. Group (i) represents remnants of square-planar Te_5 units in tetragonal Ga_2Te_5 , the inset in Figure 11(b), confirmed also by bond angle distribution $B_{TeTeTe}(\theta)$. Group (ii) is similar to chains in trigonal tellurium, $P3_12$ [57], supported by a contribution at about $\theta_{TeTeTe} \approx 103^\circ$, Figure 11(b). The two groups appear as a consequence of limited thermal stability of tetragonal gallium pentatelluride and peritectic reaction $Ga_2Te_5 \rightleftharpoons Ga_2Te_3 + 2Te$ above 768 K.

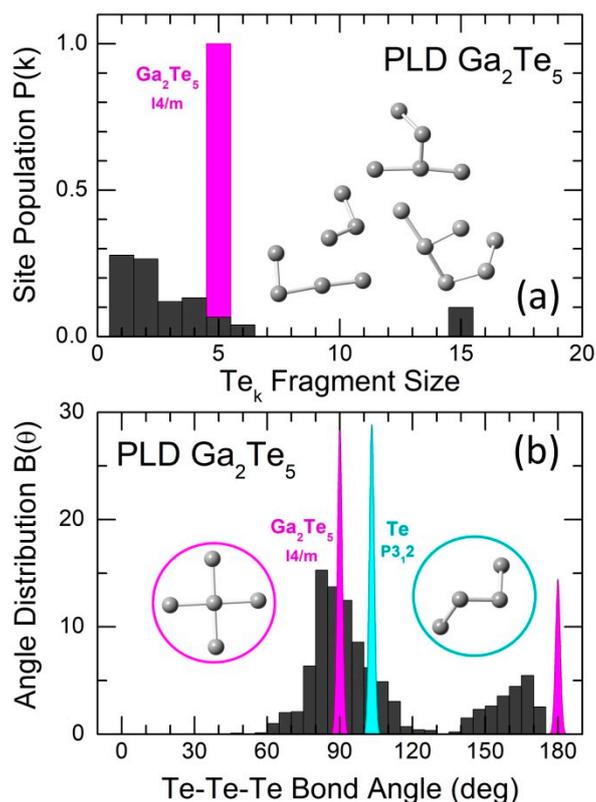


Figure 11. (a) Size and (b) Te-Te-Te bond angle distribution in Te_k oligomeric fragments, $k \leq 15$. The only population of 5-membered Te_5 square-planar fragments (crosses) in tetragonal Ga_2Te_5 is also shown in (a), as well as the characteristic $B_{TeTeTe}(\theta)$ distributions in trigonal tellurium [57] and tetragonal Ga_2Te_5 [26]. Typical Te_k fragments in glassy and crystalline pentatellurides are shown in the insets, as well as a part of the helical tellurium chain in c -Te, $P3_12$ [57].

The intermediate-range order in glassy and amorphous materials is often described by ring statistics, that is, by population of Ga_pTe_q rings in case of gallium tellurides. The ring population $R_c(p+q)$ [54] in glassy Ga_2Te_5 (this work) and Ga_2Te_3 [16] PLD films in comparison with crystalline references: tetragonal Ga_2Te_5 [26], cubic [17] and rhombohedral [85] Ga_2Te_3 is shown in Figure 12. The $R_c(p+q)$ population was found to be different for the two Ga_2Te_5 forms. The dominant 8-membered rings in tetragonal polymorph are hardly populated in the PLD film. On the contrary, the most populated $p+q = 5$ entities in g - Ga_2Te_5 are absent in c - Ga_2Te_5 . The peritectic nature of tetragonal crystal seems to be related to this difference.

Gallium sesquitelluride Ga_2Te_3 is a congruently melting compound, Figure 1(d). As it was reported earlier [16], the ring statistics in g - Ga_2Te_3 represents a disordered mixture of $R_c(p+q)$ for the ambient and high-pressure polymorphs, Figure 12(b,d), related to nanotectonic contraction in a viscous supercooled melt.

Microscopic voids and cavities in amorphous Ga-Te alloys, obtained using the Dirichlet-Voronoi tessellation [56], are displayed in Figure 13(b). The fraction of voids V_c , normalized to the volume of the FPMD simulation box, was found to be nearly invariant, $27 \leq V_c(x) \leq 29\%$, over the gallium content x between 0.20 (bulk g-GaTe₄) and 0.40 (PLD g-Ga₂Te₃). This is coherent with a small change in the number density, $\approx 2\%$ over the same composition range. Typical cavity radius varies between 0.2 and 4 Å, slightly increasing with x , Figure 13(a).

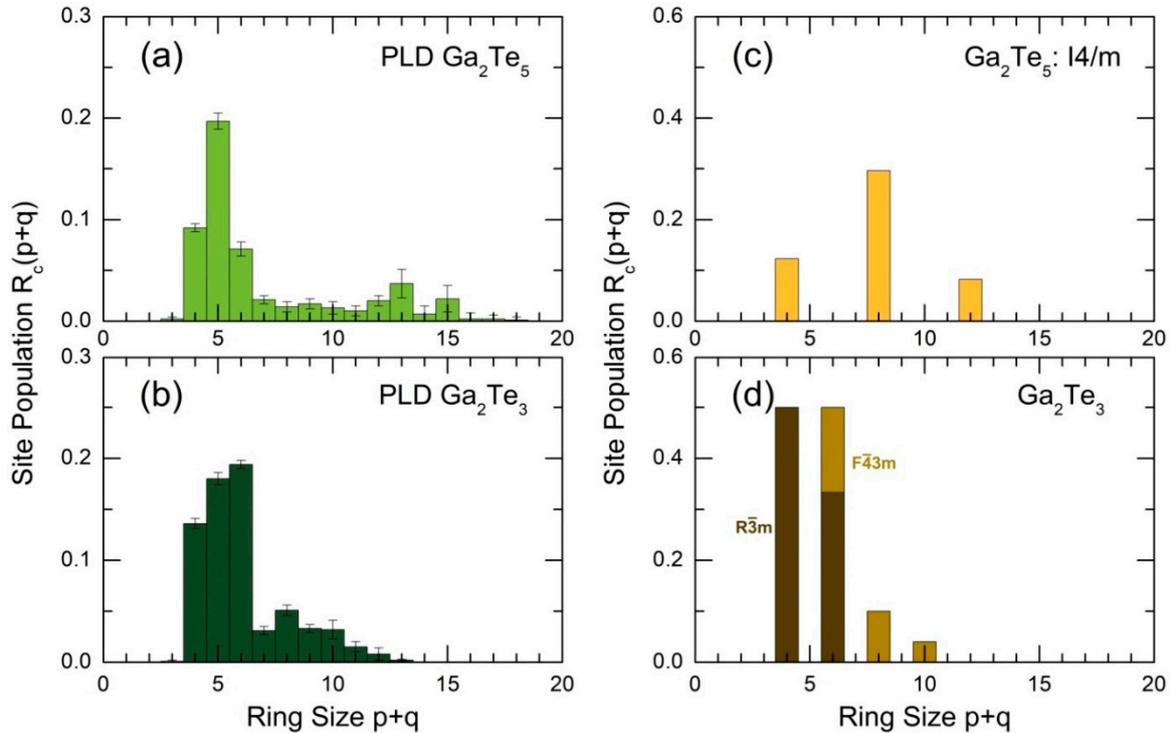


Figure 12. Ring population $R_c(p+q)$ [54] in glassy (a) Ga₂Te₅ (this work) and (b) Ga₂Te₃ [16] PLD films, and crystalline references: (c) tetragonal Ga₂Te₅ [26], (d) cubic [17] and rhombohedral [85] Ga₂Te₃.

The total electronic density of states (eDOS) is shown in Figure 14 and appears to be typical for glassy and crystalline chalcogenides [15,16,59,86-88]. The valence band (VB) consists of three sub-bands between the Fermi level E_F and -16 eV. The upper part, roughly centered at -3 eV, mostly consists of Te 5p and Ga 4p states, and also includes non-negligible d-electron contributions, as it revealed by the eDOS projections (pDOS) on Ga and Te atomic pseudo-wave functions. The middle-energy sub-band, centered at -8 eV, essentially contains Te 5p and Ga 4s electron states, while the lower part, peaked at -13 eV, has an s-character, populated by Te 5s electrons together with Ga s-, p- and d-electron contributions. The derived eDOS and pDOS are similar to those in g-Ga₂Te₃ PLD film [16] and suggest sp^3 gallium hybridization having also d-electron contribution.

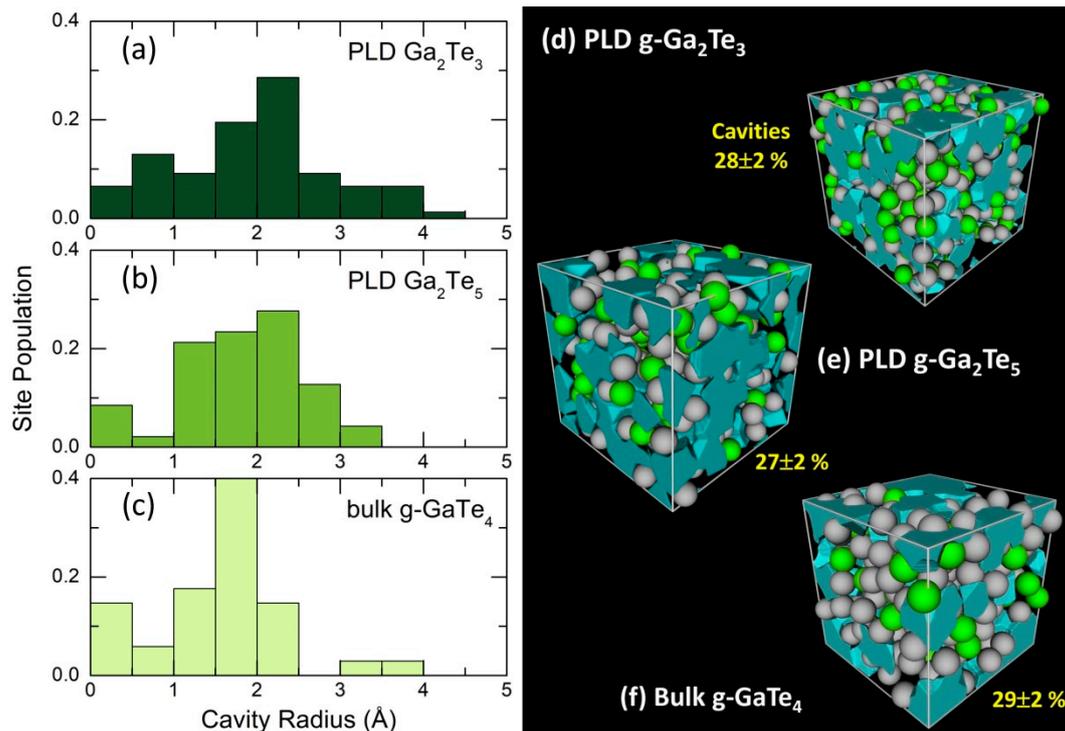


Figure 13. Microscopic voids in glassy Ga-Te alloys; typical distributions of characteristic cavity radii in (a) Ga_2Te_3 , (b) Ga_2Te_5 , and (c) GaTe_4 ; snapshots of simulation boxes with microscopic voids for (d) Ga_2Te_3 [16], (e) Ga_2Te_5 (this work) and (f) GaTe_4 [15], calculated using the Dirichlet-Voronoi approximation [56].

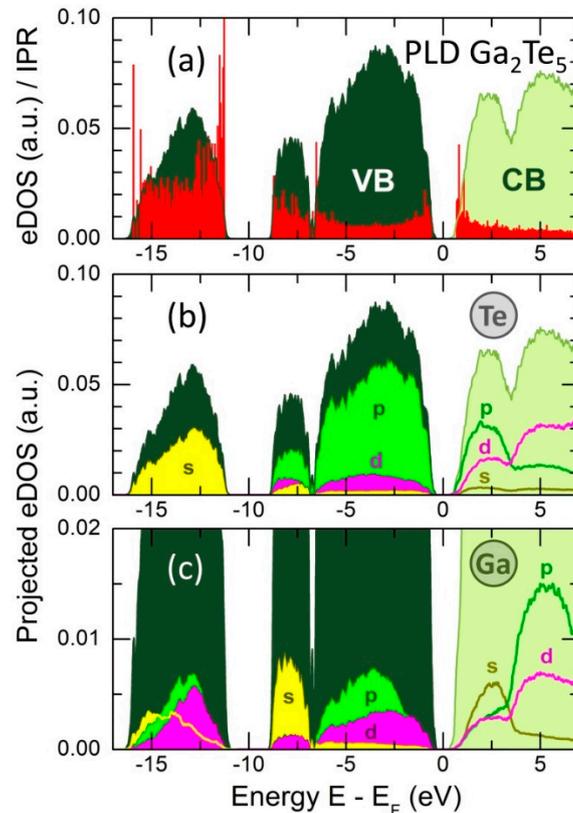


Figure 14. (a) Electronic density of states (eDOS) and inverse participation ratio (IPR, red spikes) in $\text{g-Ga}_2\text{Te}_5$ PLD film at 300 K, and projected DOS on (b) tellurium and (c) gallium atomic pseudo-wave functions. The s -, p - and d -contributions are shown. VB = valence band; CB = conduction band.

The inverse participation ratio IPR [89,90] allows localized (large IPR \rightarrow 1) and extended electron states (small IPR $\approx N^{-1}$, where N is the number of atoms in the simulation box) to be distinguished

$$\text{IPR} = \frac{\int dr |\psi(\mathbf{r})|^4}{(\int dr |\psi(\mathbf{r})|^2)^2}, \quad (14)$$

where $\psi(\mathbf{r})$ is a single-particle Kohn-Sham eigenfunction. The calculated IPRs, derived using the projections of $\psi(\mathbf{r})$ onto an atomic basis set and the atomic orbital coefficients, are shown in Figure 14(a), plotted together with the eDOS. As it was reported earlier [16,46], a higher electron localization appears at the band tails (the top of the valence and the bottom of the conduction bands), consistent with the theories of disordered semiconductors [59]. The remaining electron states in the vicinity of the bandgap are delocalized (extended). Deeper states of the lower-energy sub-bands, participating in the covalent bonding, are localized even more strongly.

The derived GGA/PBE0 bandgap $E_g^{\text{PBE0}} = 0.80$ eV appears to be smaller than the experimental optical bandgap $E_g = 0.98$ eV, Figure 3(b), but nearly identical to the interpolated electrical counterpart $2E_a = 0.82$ eV, Figure 2(a). The main contribution of the Te 5p electron states to the upper part of the valence band and at the bottom of the conduction band is also consistent with a dominant role played by Te on the electronic conductivity of Ga-Te alloys, Figure 2(b).

7. Conclusions

Pulsed laser deposition allows homogeneous and uniform Ga₂Te₅ films to be obtained, showing a distinct glass transition at 491 K and accompanied by strong crystallization peaked at 545 K. Thermal annealing of the PLD film with a DSC heating rate and below the stability limits of tetragonal Ga₂Te₅ yields a high-quality and long-living (at least, for 15 months) tetragonal polymorph, thermodynamically metastable at ambient conditions. Amorphous and crystalline Ga₂Te₅ forms show reasonably high electric contrast (two orders of magnitude at room temperature) and distinctly different optical band gap, $E_g = 0.98$ eV for g-Ga₂Te₅ and indirect optical bandgap $E_g^i = 0.40$ eV for c-Ga₂Te₅. Consequently, gallium pentatelluride can be used for memory applications as well as for photovoltaics, quantum dots-based DNA sensors and for preparation of atomically thin layers by controlled crystallization of amorphous thin films.

Tetrahedral gallium local coordination is common for the two forms; however, the intermediate-range order and tellurium subnetwork are drastically different. Square-planar Te₅ units (crosses) connecting edge-sharing Ge-Te chains in tetragonal Ga₂Te₅, which are formed by ES-GaTe₄ tetrahedra, do not survive in the glassy polymorph, leaving Te_k remnants, $3 \leq k \leq 6$, originating from the Te₅ entities, and oligomeric tellurium chains similar to those in trigonal Te. Quasi-1D edge-sharing Ga-Te chains lose their exclusive structural signature, characteristic of tetragonal polymorph, transforming into 2D and 3D structural patches of edge- and corner-sharing GaTe₄ tetrahedra. The simulated electronic density of states is consistent with experimental optical and conductivity results, and reveals a predominant role of the Te 5p electron states for electronic properties of gallium pentatelluride.

Supplementary Materials: The following supporting information can be downloaded at the website of this paper posted on Preprints.org. X-ray diffraction patterns of rapidly cooled Ga₂Te₅ melt and Ga₂Te₅ PLD film; scattering corrections for crystallized Ga₂Te₅ PLD film; FSDP parameters for glassy Ga_xTe_{1-x} alloys; bond angle distribution $B_{\text{TeGaTe}}(\theta)$ in simulated g-Ga₂Te₅.

Author Contributions: Conceptualization, E. B. and A. T.; HE-XRD, C. J. B.; synthesis, A. T., M. B. and M. K.; PLD, S. B.; FPMD, RMC and statistics, M. Kh., A. S. and D. F.; XRD, DSC and conductivity, A. T., M. B. and M. K.; optics, A. T.; paper writing, E. B. with the help of all authors.

Funding: This work was supported by the Région Hauts de France and the Ministère de l'Enseignement Supérieur et de la Recherche (CPER ECRIN), as well as by the European Fund for Regional Economic Development. Work at the Advanced Photon Source, Argonne National Laboratory, was supported in part by the Office of Basic Energy Sciences, US Department of Energy, under Contract No. DE-AC02-06CH1135. MKh work at ILIT RAS and TSU was supported by the Ministry of Science and Higher Education under agreement 075-15-2021-917; state assignment FSRC 'Crystallography and Photonics' Russian Academy of Sciences and by the Tomsk State University Development Program (Priority-2030). SB work was supported by the Estonian

Research Council under grant number PRG1023. Partly, the measurements were carried out in the resource centers of St. Petersburg State University: “Centre for X-ray Diffraction Studies” and “Centre for Thermogravimetric and Calorimetric Research”. This work was granted access to the HPC resources of IDRIS (France) under the allocation 2022-A0130910639 made by Grand Equipement National de Calcul Intensif (GENCI) and to use the CALCULCO computing platform, supported by Service Commun du Système d’Information de l’Université du Littoral Côte d’Opale (SCoSI/ULCO). The FPMD simulations were also carried out using the HPC computing resources at Lomonosov Moscow State University and at the ILIT RAS in Shatura (Moscow Region).

Data Availability Statement: The data presented in this study are available on request from the corresponding author.

Conflicts of Interest: The authors declare no conflict of interest.

References

1. Zhang, W.; Mazzarello, R.; Wuttig, M.; Ma, E. Designing crystallization in phase-change materials for universal memory and neuro-inspired computing. *Nat. Rev. Mater.* **2019**, *4*, 150–168.
2. Xu, M.; Mai, X.; Lin, J.; Zhang, W.; Li, Y.; He, Y.; Tong, H.; Hou, X.; Zhou, P.; Miao, X. Recent advances on neuromorphic devices based on chalcogenide phase-change materials. *Adv. Funct. Mater.* **2020**, *30*, 2003419.
3. Feldmann, J.; Youngblood, N.; Karpov, M.; Gehring, H.; Li, X.; Stappers, M.; Le Gallo, M.; Fu, X.; Lukashchuk, A.; Raja, A. S.; Liu, J.; Wright, C. D.; Sebastian, A.; Kippenberg, T. J.; Pernice, W. H. P.; Bhaskaran, H. Parallel convolutional processing using an integrated photonic tensor core. *Nature* **2021**, *589*, 52–61.
4. Wang, Q.; Rogers, E. T. F.; Gholipour, B.; Wang, C.-M.; Yuan, G.; Teng, J.; Zheludev, N. I. Optically reconfigurable metasurfaces and photonic devices based on phase change materials. *Nature Photon* **2016**, *10*, 60–66.
5. Shalaginov, M. Y.; An, S.; Zhang, Y.; Yang, F.; Su, P.; Liberman, V.; Chou, J. B.; Roberts, C. M.; Kang, M.; Rios, C.; Du, Q.; Fowler, C.; Agarwal, A.; Richardson, K. A.; Rivero-Baleine, C.; Zhang, H.; Hu, J.; Gu, T. Reconfigurable all-dielectric metalens with diffraction-limited performance. *Nature Commun.* **2021**, *12*, 1225.
6. Konnikova, M. R.; Khomenko, M. D.; Tverjanovich, A. S.; Bereznev, S.; Mankova, A. A.; Parashchuk, O. D.; Vasilevsky, I. S.; Ozheredov, I. A.; Shkurinov, A. P.; Bychkov, E. A. GeTe₂ phase change material for terahertz devices with reconfigurable functionalities using optical activation. *ACS Appl. Mater. Interfaces* **2023**, *15*, 9638–9648.
7. Dong, W.; Liu, H.; Behera, J. K.; Lu, L.; Ng, R. J. H.; Sreekanth, K. V.; Zhou, X.; Yang, J. K. W.; Simpson, R. E. Wide bandgap phase change material tuned visible photonics. *Adv. Funct. Mater.* **2018**, 1806181.
8. Liu, H.; Dong, W.; Wang, H.; Lu, L.; Ruan, Q.; Tan, Y. S.; Simpson, R. E.; Yang, J. K. W. Rewritable color nanoprints in antimony trisulfide films. *Sci. Adv.* **2020**, *6*, eabb7171.
9. Delaney, M.; Zeimpekis, I.; Lawson, D.; Hewak, D. W.; Muskens, O. L. A new family of ultralow loss reversible phase-change materials for photonic integrated circuits: Sb₂S₃ and Sb₂Se₃. *Adv. Funct. Mater.* **2020**, 2002447.
10. Ilie, S. T.; Faneca, J.; Zeimpekis, I.; Domínguez Bucio, T.; Grabska, K.; Hewak, D. W.; Chong, H. M. H.; Gardes, F. Y. Thermo-optic tuning of silicon nitride microring resonators with low loss non-volatile Sb₂S₃ phase change material. *Sci. Rep.* **2022**, *12*, 17815.
11. Zhu, H.; Yin, J.; Xia, Y.; Liu, Z. Ga₂Te₃ Phase change material for low-power phase change memory application. *Appl. Phys. Lett.* **2010**, *97*, 083504.
12. Lee, D.; Kim, T.; Sohn, H. Highly reliable threshold switching behavior of amorphous Ga₂Te₃ films deposited by RF sputtering. *Appl. Phys. Express* **2019**, *12*, 085504.
13. Lee, D.; Kim, T.; Kim, J.; Sohn, H. Effect of nitrogen doping on threshold voltage in amorphous Ga₂Te₃ for application of selector devices. *Phys. Status Solidi A* **2020**, *217*, 2000478.
14. Kolobov, A. V.; Fons, P.; Krbal, M.; Mitrofanov, K.; Tominaga, J.; Uruga, T. Local structure of the crystalline and amorphous states of Ga₂Te₃ phase-change alloy without resonant bonding: A combined X-ray absorption and ab initio study. *Phys. Rev. B* **2017**, *95*, 054114.
15. Bokova, M.; Tverjanovich, A.; Benmore, C. J.; Fontanari, D.; Sokolov, A.; Khomenko, M.; Kassem, M.; Ozheredov, I.; Bychkov, E. Unraveling the atomic structure of bulk binary Ga-Te glasses with surprising nanotectonic features for phase-change memory applications. *ACS Appl. Mater. Interfaces* **2021**, *13*, 37363–37379.
16. Tverjanovich, A.; Khomenko, M.; Benmore, C. J.; Bereznev, S.; Sokolov, A.; Fontanari, D.; Kiselev, A.; Lotin, A.; Bychkov, E. Atypical phase-change alloy Ga₂Te₃: Atomic structure, incipient nanotectonic nuclei, and multilevel writing. *J. Mater. Chem. C* **2021**, *9*, 17019–17032.
17. Guymont, M.; Tomas, A.; Guittard, M. The structure of Ga₂Te₃. An X-ray and high-resolution electron microscopy study. *Philos. Mag. A* **1992**, *66*, 133–139.

18. Glazov, V. M.; Chizhevskaya, S. N.; Glagoleva, N. N. *Liquid Semiconductors*; Plenum: New York, 1969; pp 142–150.
19. Chizhevskaya, S. N.; Glazov, V. M. A study of chemical interactions between gallium and tellurium in molten state. *Zh. Neorg. Khim.* **1962**, *7*, 1933–1937.
20. Ito, K.; Moynihan, C. T.; Angell, C. A. Thermodynamic determination of fragility in liquids and a fragile-to-strong liquid transition in water. *Nature* **1999**, *398*, 492–495.
21. Wei, S.; Lucas, P.; Angell, C. A. Phase change alloy viscosities down to T_g using Adam-Gibbs-equation fittings to excess entropy data: A fragile-to-strong transition. *J. Appl. Phys.* **2015**, *118*, 034903.
22. Orava, J.; Hewak, D. W.; Greer, A. L. Fragile-to-strong crossover in supercooled liquid Ag-In-Sb-Te studied by ultrafast calorimetry. *Adv. Funct. Mater.* **2015**, *25*, 4851–4858.
23. Zhu, W.; Gulbitten, O.; Aitken, B.; Sen, S. Viscosity, enthalpy relaxation and liquid-liquid transition of the eutectic liquid $\text{Ge}_{15}\text{Te}_{85}$. *J. Non-Cryst. Solids* **2021**, *554*, 120601.
24. Kassem, M.; Benmore, C. J.; Usuki, T.; Ohara, K.; Tverjanovich, A.; Bokova, M.; Brazhkin, V. V.; Bychkov, E. Transient mesoscopic immiscibility, viscosity anomaly and high internal pressure at the semiconductor-metal transition in liquid Ga_2Te_3 . *J. Phys. Chem. Lett.* **2022**, *13*, 10843–10850.
25. Taylor, G. I. The viscosity of a fluid containing small drops of another fluid. *Proc. R. Soc. London, Ser. A* **1932**, *138*, 41–48.
26. Julien-Pouzol, M.; Jaulmes, S. Alapini, F. Tellurure de gallium. *Acta Crystallogr. B* **1977**, *33*, 2270–2272.
27. Deiseroth, H. J.; Amann, P. Die Pentatelluride M_2Te_5 (M = Al, Ga, In): Polymorphie, Struktur-beziehungen und Homogenitätsbereiche. *Z. Anorg. Allg. Chem.* **1996**, *622*, 985–993.
28. Alapini, F.; Flahaut, J.; Guittard, M.; Jaulmes, S.; Julien-Pouzol, M. Système gallium-tellure. Diagramme de phases, étude structurale de GaTe , Ga_2Te_5 et de $\text{Ga}_6\text{SnTe}_{10}$. *J. Solid State Chem.* **1979**, *28*, 309–319.
29. Blachnik, R.; Irle, E. Das System Gallium-Tellur. *J. Less-Common Metals* **1985**, *113*, L1–L3.
30. Oh, C.-S.; Lee, D. N. Thermodynamic assessment of the Ga-Te system. *CALPHAD* **1992**, *16*, 317–330.
31. Ogugua, S. N.; Ntwaeaborwa, O. M.; Swart, H. C. Latest development on pulsed laser deposited thin films for advanced luminescence applications. *Coatings* **2020**, *10*, 1078.
32. Aydoğar, Ş.; Karacali, T.; Yoğurtçu, Y. K. Investigation of the switching phenomena in Ga_2Te_3 single crystals. *J. Cryst. Growth* **2005**, *279*, 110–113.
33. Fu, H.; Ying, P.; Cui, J.; Zhang, X.; Yan, Y. Thermoelectric properties of p-type Ga_2Te_3 based compounds. *Rare Metal Mater. Eng.* **2012**, *41*, 767–771.
34. Fuku, X.; Baker, P.; Iwuoha, E. Influence of quantum dot surface on electrochemical DNA sensing mechanism. *ChemElectroChem* **2020**, *7*, 770–781.
35. Siddique, S.; Gowda, C. C.; Tromer, R.; Demiss, S.; Singh Gautam, A. R.; Femi, O. E.; Kumbhakar, P.; Galvao, D. S.; Chandra, A.; Tiwary, C. S. Scalable synthesis of atomically thin gallium telluride nanosheets for supercapacitor applications. *ACS Appl. Nano Mater.* **2021**, *4*, 4829–4838.
36. Hoang Huy, V. P.; Kim, I. T.; Hur, J. Gallium-telluride-based composite as promising lithium storage material. *Nanomaterials* **2022**, *12*, 3362.
37. Guan, J.; Sun, C.; Zhang, C.; Guan, Q.; Kan, E. First-principles calculations of monolayered Al_2Te_3 : a promising 2D donor semiconductor with ultrahigh visible light harvesting. *Nanoscale* **2023**, *15*, 2578–2585.
38. Tverjanovich, A.; Khomenko, M.; Bereznev, S.; Fontanari, D.; Sokolov, A.; Usuki, T.; Ohara, K.; Le Coq, D.; Masselin, P.; Bychkov, E. Glassy GaS: Transparent and unusually rigid thin films for visible to mid-IR memory applications. *Phys. Chem. Chem. Phys.* **2020**, *22*, 25560–25573.
39. Hammersley, A. P.; Svensson, S. O.; Hanfland, M.; Fitch, A. N.; Häusermann, D. Two-dimensional detector software: from real detector to idealised image or two-theta scan. *High Pressure Res.* **1996**, *14*, 235–248.
40. Skinner, L. B.; Benmore, C. J.; Parise, J. B. Area detector corrections for high quality synchrotron X-ray structure factor measurements. *Nucl. Instrum. Methods Phys. Res.* **2012**, *662*, 61–70.
41. Kühnle, T. D.; Iannuzzi, M.; Del Ben, M.; Rybkin, V. V.; Seewald, P.; Stein, F.; Laino, T.; Khaliullin, R. Z.; Schütt, O.; Schiffmann, F.; Golze, D.; Wilhelm, J.; Chulkov, S.; Bani-Hashemian, M. H.; Weber, V.; Borschnik, U.; TAILLEFUMIER, M.; Jakobovits, A. S.; Lazzaro, A.; Pabst, H.; Müller, T.; Schade, R.; Guidon, M.; Andermatt, S.; Holmberg, N.; Schenter, G. K.; Hehn, A.; Bussy, A.; Belleflamme, F.; Tabacchi, G.; Glöb, A.; Lass, M.; Bethune, I.; Mundy, C. J.; Plessl, C.; Watkins, M.; VandeVondele, J.; Krack, M.; Hutter, J. CP2K: An electronic structure and molecular dynamics software package – Quickstep: Efficient and accurate electronic structure calculations. *J. Chem. Phys.* **2020**, *152*, 194103.
42. Perdew, J. P.; Ernzerhof, M.; Burke, K. Rationale for mixing exact exchange with density functional approximations. *J. Chem. Phys.* **1996**, *105*, 9982–9985.
43. Adamo, C.; Barone, V. Toward reliable density functional methods without adjustable parameters: The PBE0 model. *J. Chem. Phys.* **1999**, *110*, 6158–6170.
44. Tverjanovich, A.; Khomenko, M.; Benmore, C. J.; Bokova, M.; Sokolov, A.; Fontanari, D.; Kassem, M.; Usuki, T.; Bychkov, E. Bulk glassy GeTe_2 : A missing member of the tetrahedral GeX_2 family and a precursor for the next generation of phase-change materials. *Chem. Mater.* **2021**, *33*, 1031–1045.

45. Kassem, M.; Bounazef, T.; Sokolov, A.; Bokova, M.; Fontanari, D.; Hannon, A. C.; Alekseev, I.; Bychkov, E. Deciphering fast ion transport in glasses: a case study of sodium and silver vitreous sulfides. *Inorg. Chem.* **2022**, *61*, 12870–12885.
46. Kassem, M.; Benmore, C. J.; Tverjanovich, A.; Usuki, T.; Khomenko, M.; Fontanari, D.; Sokolov, A.; Ohara, K.; Bokova, M.; Kohara, S.; Bychkov, E. Glassy and liquid Sb_2S_3 : insight into the structure and dynamics of a promising functional material. *J. Mater. Chem. C* **2023**, *11*, 4654–4673.
47. Grimme, S.; Ehrlich, S.; Goerigk, L. Effect of the damping function in dispersion corrected density functional theory. *J. Comput. Chem.* **2011**, *32*, 1456–1465.
48. Micoulaut, M. Communication: Van der Waals corrections for an improved structural description of telluride based materials. *J. Chem. Phys.* **2013**, *138*, 061103.
49. Bouzid, A.; Massobrio, C.; Boero, M.; Ori, G.; Sykina, K.; Furet, E. Role of the van der Waals interactions and impact of the exchange-correlation functional in determining the structure of glassy GeTe_4 . *Phys. Rev. B* **2015**, *92*, 134208.
50. Gereben, O.; Pusztai, L. RMC_POT, a computer code for reverse Monte Carlo modeling the structure of disordered systems containing molecules of arbitrary complexity. *J. Comput. Chem.* **2012**, *33*, 2285–2291.
51. Hartwigsen, C.; Goedecker, S.; Hutter, J. Relativistic separable dual-space Gaussian pseudo-potentials from H to Rn. *Phys. Rev. B* **1998**, *58*, 3641.
52. Nosé, S. A Molecular dynamics method for simulations in the canonical ensemble. *Mol. Phys.* **1984**, *52*, 255–268.
53. Hoover, W. G. Canonical dynamics: Equilibrium phase-space distributions. *Phys. Rev. A* **1985**, *31*, 1695.
54. Le Roux, S.; Jund, P. Ring statistics analysis of topological networks: New approach and application to amorphous GeS_2 and SiO_2 systems. *Comput. Mater. Sci.* **2010**, *49*, 70–83.
55. Kohara, S.; Ohno, H.; Takata, M.; Usuki, T.; Morita, H.; Suzuya, K.; Akola, J.; Pusztai, L. Lead silicate glasses: Binary network-former glasses with large amounts of free volume. *Phys. Rev. B* **2010**, *82*, 134209.
56. Heimbach, I.; Rhiem, F.; Beule, F.; Knodt, D.; Heinen, J.; Jones, R. O. pyMolDyn: Identification, structure, and properties of cavities/vacancies in condensed matter and molecules. *J. Comput. Chem.* **2017**, *38*, 389–394.
57. Cherin, P.; Unger, P. Two-dimensional refinement of the crystal structure of tellurium. *Acta Crystallogr.* **1967**, *23*, 670–671.
58. Nassary, M. M.; Gerges, M. K.; Shaban, H. T.; Salwa, A. S. Some physical properties of Ga_2Te_3 single crystals. *Physica B* **2003**, *337*, 130–137.
59. Mott, N. F.; Davis, E. A. *Electronic Processes in Non-Crystalline Materials*, 2nd ed.; Clarendon Press: Oxford, 1979; pp 5–64, 491–497.
60. Ray, A. K.; Swan, R.; Hogarth, C. A. Conduction mechanisms in amorphous tellurium films. *J. Non-Cryst. Solids* **1994**, *168*, 150–156.
61. Bekheet, A. E. Electrical and optical properties of amorphous Ga_2Te_3 films. *Eur. Phys. J.: Appl. Phys.* **2001**, *16*, 187–193.
62. Swanepoel, R. Determination of the thickness and optical constants of amorphous silicon. *J. Phys. E: Sci. Instrum.* **1983**, *16*, 1214–1222.
63. Jenkins, F. A.; White, H. E. *Fundamentals of Optics*, 4th ed.; McGraw-Hill: New York, 2001; pp. 479–485.
64. Tauc, J. Optical properties of amorphous semiconductors. In *Amorphous and Liquid Semiconductors*, ed. J. Tauc; Plenum: London, 1974; pp. 159–220.
65. Heller, W.; Bhatnagar, H. L.; Nakagaki, M. Theoretical investigations on the light scattering of spheres. XIII. The "wavelength exponent" of differential turbidity spectra. *J. Chem. Phys.* **1962**, *36*, 1163–1170.
66. Melik, D. H.; Fogler, H. S. Turbidimetric determination of particle size distributions of colloidal systems. *J. Colloid Interface Sci.* **1983**, *92*, 161–180.
67. Green, M. A.; Keevers, M. J. Optical properties of intrinsic silicon at 300 K. *Prog. Photovolt.: Res. Appl.* **1995**, *3*, 189–192.
68. Zacharias M.; Feliciano Giustino, F. One-shot calculation of temperature-dependent optical spectra and phonon-induced band-gap renormalization. *Phys. Rev. B* **2016**, *94*, 075125.
69. Koughia, K.; Singh, J.; Kasap, S. O.; Ruda, H. E. Fundamental optical properties of materials II. In *Optical Properties of Condensed Matter and Applications*, ed. J. Singh; Wiley: Chichester, UK, 2006; pp. 27–46.
70. Noffsinger, J.; Kioupakis, E.; Van de Walle, C. G.; Louie, S. G.; Cohen, M. L. Phonon-assisted optical absorption in silicon from first principles. *Phys. Rev. Lett.* **2012**, *108*, 167402.
71. Zacharias, M.; Patrick, C. E.; Giustino, F. Stochastic approach to phonon-assisted optical absorption. *Phys. Rev. Lett.* **2015**, *115*, 177401.
72. Kangsabanik, J.; Svendsen, M. K.; Taghizadeh, A.; Crovetto, A.; Thygesen, K. S. Indirect band gap semiconductors for thin-film photovoltaics: High-throughput calculation of phonon-assisted absorption. *J. Am. Chem. Soc.* **2022**, *144*, 19872–19883.
73. Bullett, D. W. Square-planar coordinated tellurides – The electronic structures of Re_2Te_5 , Ga_2Te_5 and K_2SnTe_5 . *Solid State Commun.* **1984**, *51*, 51–53.

74. Bychkov, E.; Benmore, C. J.; Price, D. L. Compositional changes of the first sharp diffraction peak in binary selenide glasses. *Phys. Rev. B* **2005**, *72*, 172107.
75. Bychkov, A.; Cuello, G. J.; Kohara, S.; Benmore, C. J.; Price, D. L.; Bychkov, E. Unraveling the atomic structure of Ge-rich sulfide glasses. *Phys. Chem. Chem. Phys.* **2013**, *15*, 8487–8494.
76. Hannon, A. C. *XTAL: a Program for Calculating Interatomic Distances and Coordination Numbers for Model Structures*; Rutherford-Appleton Laboratory Report RAL-93-063, 1993. <http://www.wisis2.isis.rl.ac.uk/disordered/ACH/Software/xtal.htm>.
77. Voleská, I.; Akola, J.; Jónvári, P.; Gutwirth, J.; Wágner, T.; Vasileiadis, T.; Yannopoulos, S. N.; Jones, R. O. Structure, electronic, and vibrational properties of glassy $\text{Ga}_{11}\text{Ge}_{11}\text{Te}_8$: Experimentally constrained density functional study. *Phys. Rev. B* **2012**, *86*, 094108.
78. Pethes, I.; Piarristeguy, A.; Pradel, A.; Michalik, S.; Nemausat, R.; Darpentigny, J.; Jónvári, P. Short range order and topology of $\text{Ge}_x\text{Ga}_x\text{Te}_{100-2x}$ glasses. *J. Alloys Compd.* **2020**, *834*, 155097.
79. Ichikawa, T. Electron diffraction study of the local atomic arrangement in amorphous tellurium films. *Phys. Stat. Solidi B* **1973**, *56*, 707–715.
80. Brazhkin, V. V.; Bychkov, E.; Tsiok, O. B. As_2Te_3 glass under high hydrostatic pressure: Polyamorphism, relaxation, and metallization. *Phys. Rev. B* **2017**, *95*, 054205.
81. Chau, P.-L.; Hardwick, A. J. A new order parameter for tetrahedral configurations. *Mol. Phys.* **1998**, *93*, 511–518.
82. Errington, J. R.; Debenedetti, P. G. Relationship between structural order and the anomalies of liquid water. *Nature* **2001**, *409*, 318–321.
83. Caravati, S.; Bernasconi, M.; Kühne, T. D.; Krack, M.; Parrinello, M. Coexistence of tetrahedral- and octahedral-like sites in amorphous phase change materials. *Appl. Phys. Lett.*, **2007**, *91*, 171906.
84. Jones, C. Y.; Bryan, J. C.; Kirschbaum, K.; Edwards, J. G. Refinement of the crystal structure of digallium trisulfide, Ga_2S_3 . *Z. Kristallogr. NCS* **2001**, *216*, 327–328.
85. Serebryanaya, N. R. The crystal structure of pressure-induced phases of In_2Te_3 and Ga_2Te_3 . *Powder Diffr.* **1992**, *7*, 99–102.
86. S. Blaineau, S.; Jund, P. Electronic structure of amorphous germanium disulfide via density-functional molecular dynamics simulations. *Phys. Rev. B* **2004**, *70*, 184210.
87. J. Akola, J.; Jones, R. O. Structural phase transitions on the nanoscale: The crucial pattern in the phase-change materials $\text{Ge}_2\text{Sb}_2\text{Te}_5$ and GeTe . *Phys. Rev. B* **2007**, *76*, 235201.
88. Dragoni, D.; Gabardi, S.; Bernasconi, M. First-principles study of the liquid and amorphous phases of In_2Te_3 . *Phys. Rev. Mater.*, **2017**, *1*, 035603.
89. Kalikka, J.; Akola, J.; Jones, R. O.; Kohara, S.; Usuki, T. Amorphous $\text{Ge}_{15}\text{Te}_{85}$: density functional, high-energy X-ray and neutron diffraction study. *J. Phys.: Condens. Matter* **2012**, *24*, 015802.
90. Caravati, S.; Bernasconi, M.; Parrinello, M. First-principles study of liquid and amorphous Sb_2Te_3 . *Phys. Rev. B* **2010**, *81*, 014201.

Disclaimer/Publisher's Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.