

Preparation and dielectric properties of BaBi 1.8 Ln 0.2 Nb 2 O 9 (Ln = Ce, Gd) ceramics

Mohamed Afqir, Amina Tachafine, Didier Fasquelle, Mohamed Elaatmani,

Jean-Claude Carru, Abdelouahad Zegzouti, Mohamed Daoud

► To cite this version:

Mohamed Afqir, Amina Tachafine, Didier Fasquelle, Mohamed Elaatmani, Jean-Claude Carru, et al.. Preparation and dielectric properties of BaBi 1.8 Ln 0.2 Nb 2 O 9 (Ln = Ce, Gd) ceramics. Materials Science-Poland, 2018, 36 (1), pp.46-50. 10.1515/msp-2018-0027 . hal-04465484

HAL Id: hal-04465484 https://ulco.hal.science/hal-04465484v1

Submitted on 22 Feb 2024

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial - NoDerivatives 4.0 International License

Preparation and dielectric properties of $BaBi_{1.8}Ln_{0.2}Nb_2O_9$ (Ln = Ce, Gd) ceramics

Mohamed Afqir^{1,2,*}, Amina Tachafine², Didier Fasquelle², Mohamed Elaatmani¹, Jean-Claude Carru², Abdelouahad Zegzouti¹, Mohamed Daoud¹

¹ Laboratoire des Sciences des Matériaux Inorganiques et leurs Applications, Faculté des Sciences Semlalia, Université Cadi Ayyad, Marrakech, Maroc

² Unité de Dynamique et Structure des Matériaux Moléculaires, Université du Littoral- Côte d'Opale, Calais, France

The main subject of the presented research is to investigate the dielectric properties of $BaBi_{1.8}Ln_{0.2}Nb_2O_9$ (Ln = Ce, Gd) ceramics prepared by conventional solid state reaction route. The materials were examined using XRD and FT-IR methods. Moreover, the AC conductivity, dielectric constant and dielectric loss of the ceramics were determined. X-ray diffraction confirmed that all these compounds crystallize in an orthorhombic structure. Fourier transform infrared spectroscopy study confirmed the presence of two characteristic vibration bands located at around 617 cm⁻¹ and 818 cm⁻¹ for BaBi₂Nb₂O₉. The experimental results show that the substitution of Bi by Ce or Gd causes a decrease in Curie temperature, dielectric constant and dielectric loss.

Keywords: BaBi2Nb2O9; structure; dielctrics

1. Introduction

Aurivillius bismuth-layer structured ferroelectrics (BLSFs), owing to their excellent dielectric and piezoelectric properties, are attractive for a wide range of applications such as ferroelectric non-volatile random access memory (FeRAM) storage devices over a wide range of temperatures [1]. BaBi₂Nb₂O₉, SrBi₂Nb₂O₉ and SrBi₂Ta₂O₉ are several bismuth-layer compounds which suffer from high dielectric loss, due to volatilization of bismuth. To solve more or less this problem, many researches proposed to substitute bismuth by lanthanide [2-4], what has been presented in our earlier paper [5]. SrBi₂Nb₂O₉ and SrBi₂Ta₂O₉ are superior candidates for new generation of ferroelectric random-access memories (FeRAM) in comparison to $BaBi_2Nb_2O_9$ [6]. To the best of our knowledge, there has been lack of sufficient information about rare-earthdoped BaBi₂Nb₂O₉ ceramics. In this paper, BaBi_{1.8}Ln_{0.2}Nb₂O₉ (Ln: Ce and Gd) samples were

synthesized through a solid-state reaction method. In addition, structural and dielectric responses were investigated.

2. Experimental

Synthesis of the materials was performed starting from stoichiometric amounts of Bi₂O₃ (Rectapur, 99 %), BaCO₃(Acros Organics, 90 %), Nb₂O₅ (Acros Organics, 99.5 %), Ce₂O₃, and Gd₂O₃ (Alfa Aesar, 99.99 %). The stoichiometric amounts of the raw materials were thoroughly ground manually in a mortar for one hour and then calcined for 12 hours at 1100 °C. The samples were then uniaxially pressed at 2 MPa into pellets having 6 mm in diameter. The pellets were covered with powders of the same materials and heated for 8 hours at 1150 °C. The powder ceramics were characterized by X-ray diffraction with a powder diffractometer (PANalytical/Philips X'Pert) and infrared spectroscope (KBr-pellet/Bruker, Vertex 70 DTGS). For dielectric measurements, both the surfaces of pellets were painted uniformly with a silver conductive paste (Alfa Aesar) and cured for 1 hour

^{*}E-mail: mohamed.afqir@yahoo.fr

^{© 2018.} This is an open access article distributed under the Creative Commons Attribution-NonCommercial-NoDerivatives 3.0 License. (http://creativecommons.org/licenses/by-nc-nd/3.0/)

at 400 °C. The temperature dependence of dielectric properties was measured at temperature ranging from room temperature to 500 °C using an LCR meter HP 4284A (Hewlett Packard Co).

3. Result and discussion

Fig. 1 shows X-ray diffraction patterns of pure and doped BaBi₂Nb₂O₉. According to the standard pattern of BaBi₂Nb₂O₉ (PDF #00-049-0607), all the observed reflections could be indexed to orthorhombic structure. The highest peak corresponding to $(1 \ 1 \ 5)$ orientation is consistent with the $(1\ 1\ 2m+1)$ highest diffraction peak in bismuthlayered structured ferroelectric systems [7]. Table 1 shows the effect of Ce and Gd on the crystal structure. The values a, b, c, V and their experimental error have been calculated using the UnitCell program. It can be seen that the values of c are much higher than the values of a and b which implies that the crystal structure is anisotropic. It can be seen that the unit cell volume decreases with a decrease in the ionic radius (ionic radii of Gd^{3+} , Ce^{3+} and Bi³⁺ are 0.938 Å, 1.034 Å and 1.02 Å, respectively), indicating that these cations have diffused into the BaBi₂Nb₂O₉ lattice.



Fig. 1. XRD patterns of the Ln-doped BaSrBi₂Nb₂O₉ ceramic powders.

FT-IR transmittance spectra of these powders are shown in Fig. 2. A broad band around ~ 3442 cm⁻¹ is assigned to the O–H stretching vibrations of adsorbed water molecules present in the sample. Around ~2933 cm⁻¹ and ~1625 cm⁻¹ the band is assigned to the C–H stretching and bending vibrations, respectively. A band at ~2360 cm⁻¹ relates to CO₂. The bands at 1420 cm⁻¹, 1060 cm⁻¹ are assigned to carbonate. Two bands are noted around ~617 cm⁻¹ and ~818 cm⁻¹ which are respectively assigned to the symmetric and asymmetric stretching of the Nb–O bond in the NbO₆ octahedral. These two bands are characteristic of niobates ABi₂Nb₂O₉ (A = Sr, Ca, Ba, Pb) [8–10]. It is found that FT-IR spectra of this species show a strong resemblance. Thus, Ce or Gd substitution will not have a significant enough effect on the shifts in frequencies.



Fig. 2. FT-IR spectra of Ln-doped BaBi₂Nb₂O₉ ceramic powders.

Fig. 3 shows dielectric constant ϵ' and loss tangent tan δ of Ce and Gd-doped BaBi₂Nb₂O₉ ceramics as a function of temperature at measurement frequencies of 10² Hz, 10³ Hz and 10⁴ Hz. The relationship between the real ϵ' and imaginary ϵ'' parts of the complex dielectric constant and loss tangent tan δ are expressed by the following relations:

$$\varepsilon' = \frac{Cd}{A\varepsilon_0} \tag{1}$$

$$\varepsilon'' = \varepsilon' \tan \delta$$
 (2)

where C is the capacitance, d is the pellet thickness, A is the electrode area, and ϵ_0 is the permittivity of free space ($\epsilon_0 = 8.85 \times 10^{-12}$ F/m).

	BBN	BBCeN	BBGdN
a [Å]	$5.602{\pm}0.00108$	5.595±0.00108	5.551±0.00108
b [Å]	$5.630{\pm}0.00028$	$5.67{\pm}0.00028$	$5.562{\pm}0.00028$
c [Å]	$25.481{\pm}0.00174$	$25.556{\pm}0.00174$	$25.669{\pm}0.00174$
V [Å] ³	$803.652{\pm}0.1411$	$810.730{\pm}0.1412$	$792.522{\pm}0.1412$

Table 1. Lattice parameters of Ce and Gd-doped BaBi₂Nb₂O₉.

The Curie temperatures (T_C) have been found at about 340 °C, 330 °C and 310 °C for BBN, BBGdN and BBCeN, respectively. The peak value of dielectric constant (ϵ'_{max}) is 1544 for BBN, which is higher than that of BBGdN (228) and BB-CeN (485), when measured at 1 kHz. The loss tangent increases with an increase in the temperature and reaches to local maximum at 335 °C, 330 °C and 310 °C for BBN, BBGdN and BBCeN, respectively. This loss tangent additionally confirms the phase transition from ferroelectric to paraelectric phase at the Curie temperature. The values of tanδ at room temperature decreas from 0.1 (BBN) to ~0.06 (Ln-doped BBN) when measured at 10 kHz.

The Curie temperature obtained for the BBN ceramics is comparable to those previously reported in the literature [11]. The significant reduction of ϵ'_{max} may be explained by macroscopic Clausius-Mossotti relationship and polarizabilities as given by Shannon [12] ($\alpha_{Bi}^{3+} = 6.12 \text{ Å}^3$, $\alpha_{Gd}^{3+} = 4.37 \text{ Å}^3$, $\alpha_{Ce}^{3+} = 6.15 \text{ Å}^3$).

Fig. 4 shows Curie-Weiss fittings of dielectric constant at 1 kHz according to equation 3:

$$\varepsilon' = \frac{C}{T - T_{CW}} \tag{3}$$

where C is the Curie constant and T_{CW} is the Curie-Weiss temperature, which differs from the Curie temperature. The magnitude of the Curie constant is of about 10⁵ °C, indicating the displacive nature of the observed transition. The ratio of the Curie constants above and below the phase transition is equal to 1 for SBCeN, which is of the same order as for SBGdN, but smaller than BBN (~1.3), referring to the first-order ferroelectric phase transition.

Fig. 5 shows the dependence of AC conductivity on inverse of temperature at 1 kHz. The curve was fitted with Arrhenius equation:

$$\sigma = \omega \varepsilon_0 \varepsilon' \tan \delta = \sigma_0 e^{-\frac{La}{k_B T}} \tag{4}$$

where E_a is the activation energy, k_B is the Boltzmann constant and σ_0 is pre-exponential factor. In the lower temperature region (3.5 K⁻¹ to 2.5 K⁻¹), conductivity shows a weak temperature dependence. In the higher temperature region (2.5 K⁻¹ to 1 K⁻¹), conductivity shows a strong temperature dependence.

The activation energy calculated from the slope of the curve in the inset (Fig. 5) is found to be 1.06 eV, 0.95 eV and 0.69 eV for BBN, BBCeN and BBGdN, respectively, when measured at 1 kHz. A decrease in activation energy may be attributed to the presence of Ce³⁺ and Gd³⁺ ions in the lattice, which reduce the oxygen vacancy formation. The bond dissociation energy of Gd–O (353 kJ/mol) Ce–O (353 kJ/mol) is larger than that of Bi–O (343 kJ/mol) [12, 13]. Hence, the substitution of Bi by Ce or Gd suppresses the formation of oxygen vacancies.

In the light of the above mentioned, BBN ceramics shows high values of $\tan \delta$, what restricts its practical applications. On the other hand, the Ce and Gd-doped BBN ceramics show dielectric properties, which are desirable in nonvolatile FeRAM applications.

4. Conclusions

 $BaBi_{1.8}Ln_{0.2}Nb_2O_9$ (Ln = Ce, Gd) samples have been prepared by the solid-state reaction method. The XRD patterns showed the presence of the parent $BaBi_2Nb_2O_9$ phase. The difference observed in unit cell parameters has been correlated with the ionic radius. The appearance of Nb–O vibration bands in FT-IR spectra is characteristic of barium bismuth niobate phase formation.



Fig. 3. Temperature dependence of dielectric constant ϵ' and loss tangent tanb of Ln-doped BaBi₂Nb₂O₉ ceramics.

BBGdN respectively. The and

Dielectric measurements revealed a decrease BaBi2Nb2O9 are higher than those of Ce and in the Curie temperature from 340 °C for Gd-doped BaBi₂Nb₂O₉ ceramics. Moreover, BBN to 330 °C and 310 °C for BBCeN the reduction in the dielectric loss tand may be dielectric attributed to suppression of oxygen vacancies constant ϵ' and loss tangent tan δ of pure by the cationic vacancies created by cerium and



Fig. 4. Temperature dependence of reciprocal dielectric constant.



Fig. 5. Conductivity in an Arrhenius plot for Ln-doped BBN at 10 kHz. The inset shows the variation of the conductivity data fitting in the 95 $^{\circ}$ C to 105 $^{\circ}$ C temperature range.

gadolinium doping. These results are useful for the potential applications of $BaBi_{1.8}Ln_{0.2}Nb_2O_9$ (Ln = Ce, Gd) ceramics.

References

- [1] PARK B.H., KANG B.S., BU S.D., NOH T.W., LEE J., JO W., *Nature.*, 401 (1999), 682.
- [2] SHRIVASTAVA V., JHA A.K., MENDIRATTA R.G., *Mater. Lett.*, 60 (2006), 1459.
- [3] SUN L., CHU J.-H., YANG P.-X., YUE F.-Y., LIU Y.-W., FU C.-D., MAO C.-L., *Trans. Nanoferrous Met. Soc. China.*, 19 (6) (2009), 1459.
- [4] ZOU H., YU Y., LI J., CAO Q., WANG X., HOU J., Mater. Res. Bull., 69 (2015), 112.
- [5] AFQIR M., TACHAFINE A., FASQUELLE D., ELAAT-MANI M., CARRU J., ZEGZOUTI A., DAOUD M., Process. Appl. Ceram., 3 (2016), 183.
- [6] LAHA A., KRUPANIDHI S.B., J. Appl. Phys., 92 (2002), 415.
- [7] SARAH P., Procedia Eng., 10 (2011), 2684.
- [8] KASPROWICZ D., LAPINSKI A., RUNKA T., SPEGH-INI A., BETTINELLI M., J. Alloy. Compd., 478 (2009), 30.
- [9] ROCHA DA M.J.S., FILHO M.C.C., THEOPHILO K.R.B., DENARDIN J.C., VASCONCELOS DE I.F., ARAUJO DE E.B., SOMBRA A.S.B., *Mater. Sci. Appl.*, 3 (1) (2012), 6.
- [10] RAMARAGHAVULU R., BUDDHUDU S., Ferroelectrics., 460 (2014), 57.
- [11] ADAMCZYK M., UJMA Z., PAWE M., SZYMCZAK L., J. Mater. Sci., (2006), 79 (2006) 435.
- [12] SHANNON R.D., J. Appl. Phys., 73 (1993) 438.
- [13] WU Y., CAO G., Mater. Res. Soc., 15 (2000), 1583.
- [14] COTTRELL T.L., *The Strengths of Chemical Bond*, Hardcover, 1958.

Received 2017-02-25 Accepted 2018-02-17